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| SIXTH EDITION |

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 $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$
 $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF} + \text{Cl}_2$
 $2\text{H}_2\text{O} + 2\text{H}_2 \rightarrow 2\text{H}_2\text{O} + 5\text{H}_2$
 $1\text{Pb} + \text{OH} \rightarrow 2\text{HCl} + 2\text{H}_2\text{O} + 1\text{PbCl}_2$
 $2\text{Al} + 3\text{K}_2\text{SO}_4 \rightarrow 4\text{KBr} + 1\text{Al}_2\text{SO}_4 + 3\text{H}_2$
 $4\text{CH}_4 + 2\text{O}_2 \rightarrow 1\text{CO}_2 + 2\text{H}_2\text{O} + 8\text{H}_2\text{O} + 1\text{C}_3\text{H}_8 + 5\text{O}_2 + 3\text{CO}_2 + 4\text{H}_2$

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 $1\text{O}_2 + 2\text{NH}_3 \rightarrow 3\text{HNO}_3 + 3\text{H}_2\text{O} + 2\text{O}_2 + 3\text{H}_2$
 $2\text{O}_2 + 2\text{H}_2\text{O} + 2\text{H}_2\text{O} + 4\text{RbH}_2\text{O} + \text{RbOH} + 2\text{H}_2 + 5\text{Cl}_2 + 2\text{KBr} + \text{KCl} + \text{Br}_2 + 6\text{N}_2 + 2\text{H}_2 + \text{NH}_3 + 7\text{Ndf} + 3\text{Ca} + \text{Nd} + \text{Caf} + 2\text{H}_2 + 8\text{CH}_4 + \text{H}_2\text{O} + \text{CO}_2 + 9\text{C}_2\text{H}_5\text{OH}$

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 $\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$
 $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF} + \text{Cl}_2$
 $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 $\text{Pb} + \text{OH} \rightarrow 2\text{HCl} + 2\text{H}_2\text{O} + 1\text{PbCl}_2$
 $2\text{Al} + 3\text{K}_2\text{SO}_4 \rightarrow 6\text{KBr} + 1\text{Al}_2\text{SO}_4 + 3\text{CH}_4 + 2\text{O}_2 + 1\text{CO}_2 + 2\text{H}_2\text{O}$

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 $4\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$
 $2\text{FeBr}_3 + 3\text{H}_2\text{SO}_4 \rightarrow 1\text{Fe}_2(\text{SO}_4)_3 + 3\text{H}_2\text{O} + 2\text{C}_2\text{H}_5\text{OH}$

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