15CHE12/22

First/Second Semester B.E. Degree Examination, June/July 2016 **Engineering Chemistry**

Time: 3 hrs.

Max. Marks, 80

		Note: Answer any FIVE full questions, choosing one full question from each mod	lule.
		No.	
		Module-1	
1	21	Explain the construction and working of silver-silver chloride electrode	(05 Marks)
	b.	What are ion selective electrodes? Discuss the construction and working of a glas-	
			(06 Marks)
	C.	Explain the construction and working of Zinc-Air cell.	(05 Marks)
		OR	
2	31	Derive Nerst expression for single electrode potential	(05 Marks)
	b.	Explain the following battery characteristics ;	
		i) Energy efficiency	
		ii) Shelf life	100000000000000000000000000000000000000
		iii) Cycle life	(06 Marks)
	C.	Explain the construction and working of Methanol - Oxygen fuel cell w	nth H ₂ SO ₄
		electrolyte.	(05 Marks)
			(US Marks)
		Module-2	
3	8362	Explain electrochemical theory of corrosion with an example.	(05 Marks)
-	b.	Discuss the differential metal corrosion with an example.	(05 Marks)
	C	What is electroless plating? Mention the differences between electoplating and	
	760.10	plating.	(06 Marks)
		. 2	5 (T-000000000000000000000000000000000000
		OR	
4	28.	Discuss the following factors influencing nature of electrodeposit.	(06 Marks)
		i) Brighteners	
		(ii) Levellers and	
	-	O iii) pH	
	1 B-	Explain electroplating of Decorative chromium.	(05 Marks)
V	Ç.	Discuss the process of Galvanization of Iron sheet.	(05 Marks)
3		Module-3	
- 5	ac.	How Calorific value of a solid fuel is determined using bomb calorimeter?	(05 Marks)
	ъ.	What is meant by cracking? Describe with a neat diagram, fluidized bed cracking	
			(06 Marks)
	C.	Explain the construction and working of a photovoltaic cell.	(05 Marks)

Engineering Chemistry Paper Solved 20

Scott C. Dulebohn

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