Name: ______ Date: ______

Net Ionic Equations: Practice Problems

Balance the following equations. Show the ionic and net ionic forms.

- Mg(NO₃)₂(aq.) + Na₂CrO₄(aq.) → MgCrO₄(s) + NaNO₃(aq.) lonic:
 Net ionic:
- 2) Na₃PO₄(aq.) + Ni(ClO₄)₂(s) → Ni₃(PO₄)₂(s) + NaClO₄(aq.) lonic:
 Net ionic:
- 3) AgNO₃(aq.) + MgI₂(aq.) → AgI(s) + Mg(NO₃)₂(aq.) lonic:
 Net ionic:
- 4) Ca(OH)₂(aq.) + FeCl₃(aq.) → Fe(OH)₃(s) + CaCl₂(aq.) lonic:
 Net ionic:
- 5) RbF(aq.) + CuSO₄(aq.) → Rb₂SO₄(aq.) + CuF₂(aq.) lonic:
 Net ionic:
- 6) (NH₄)₂S(aq.) + Co(NO₃)₂(aq.) → CoS(s) + NH₄NO₃(aq.) lonic:
 Net ionic:
- 7) Na₃PO₄(aq.) + FeCl₃(aq.) → FePO₄(s) + NaCl(aq.) lonic:
 Net ionic:
- 8) AgNO₃(aq.) + HCI(aq.) → AgCI(s) + HNO₃(aq.) lonic:
 Net ionic:
- 9) MgBr₂(aq.) + KOH(aq.) → Mg(OH)₂(s) + KBr(aq.) lonic: Net ionic:
- 10) AlCl₃(aq.) + K₃PO₄(aq.) → AlPO₄(s) + KCl(aq.) lonic:
 Net ionic:

Ions Questions And Answers

Purshotam Kaushik

Ions Questions And Answers:

Ions Questions And Answers Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Ions Questions And Answers**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we will delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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