

Ionic Naming Worksheet #2

Name the following ionic compounds:

- 1) $\text{Mo}(\text{SO}_4)_3$ _____
- 2) $\text{Ti}(\text{NO}_3)_3$ _____
- 3) NaH _____
- 4) $\text{BeSO}_3 \cdot 7\text{H}_2\text{O}$ _____
- 5) $\text{Fe}_3(\text{PO}_4)_2$ _____
- 6) $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$ _____
- 7) $\text{V}(\text{H}_2\text{PO}_4)_4$ _____
- 8) $\text{Mn}_3(\text{PO}_4)_7$ _____
- 9) $\text{Fe}(\text{CN})_3$ _____
- 10) BaSO_4 _____

Write the formulas for the following ionic compounds:

- 11) vanadium (V) arsenide _____
- 12) cobalt (III) nitrite _____
- 13) yttrium perchlorate _____
- 14) silver phosphide _____
- 15) tin (IV) selenide _____
- 16) cadmium sulfite _____
- 17) nickel (III) sulfate _____
- 18) scandium hydroxide _____
- 19) copper (I) oxide _____
- 20) ammonium nitride _____

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