

## Second Semester Final Exam

**Multiple Choice:** Please mark the answer to the multiple choice questions on your bubble sheet.

- 1) The molar mass of magnesium cyanide,  $\text{Mg}(\text{CN})_2$ , is approximately
  - A) 50.3 g
  - B) 64.3 g
  - C) 76.3 g
  - D) 100.6 g
- 2) The total number of atoms in  $(\text{NH}_4)_3\text{BO}_3$  is
  - A) 12
  - B) 17
  - C) 19
- 3) What is the mass in grams of 5.90 mol  $\text{C}_8\text{H}_{18}$ ?
  - A) 0.0512 g
  - B) 389 g
  - C) 673 g
  - D) 19.4 g
- 4) How many atoms are there in 1 mole of helium?
  - A)  $6.02 \times 10^{23}$  atoms
  - B) 22.4 atoms
  - C) 4.0 atoms
  - D) 0.0821 atoms
- 5) The percent composition of silver (Ag) in  $\text{AgNO}_3$  is
  - A) 0.625 %
  - B) 1.57 %
  - C) 62.5 %
  - D) 157 %
  - E) 169.9 %
- 6) Which of the following is an empirical formula?
  - A)  $\text{C}_2\text{H}_4$
  - B)  $\text{C}_2\text{H}_6$
  - C)  $\text{CH}_4$
  - D)  $\text{C}_3\text{H}_6$
- 7) Which of the following could be a molecular formula for the empirical formula  $\text{CH}_2$ ?
  - A)  $\text{C}_3\text{H}_6$
  - B)  $\text{C}_6\text{H}_3$
  - C)  $\text{CH}$
  - D)  $\text{CH}_4$
- 8) What is the coefficient in front of the  $\text{O}_2$  when the following reaction is correctly balanced?  
$$\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$$
  - A) 1
  - B) 2
  - C) 3
  - D) 4
  - E) 5

# Ied 2nd Semester Final Exam Answers

**Lauren Gardner**



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## **Table of Contents Ied 2nd Semester Final Exam Answers**

1. Understanding the eBook Ied 2nd Semester Final Exam Answers
  - The Rise of Digital Reading Ied 2nd Semester Final Exam Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Ied 2nd Semester Final Exam Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Ied 2nd Semester Final Exam Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Ied 2nd Semester Final Exam Answers
  - Personalized Recommendations
  - Ied 2nd Semester Final Exam Answers User Reviews and Ratings
  - Ied 2nd Semester Final Exam Answers and Bestseller Lists
5. Accessing Ied 2nd Semester Final Exam Answers Free and Paid eBooks
  - Ied 2nd Semester Final Exam Answers Public Domain eBooks
  - Ied 2nd Semester Final Exam Answers eBook Subscription Services
  - Ied 2nd Semester Final Exam Answers Budget-Friendly Options

6. Navigating Ied 2nd Semester Final Exam Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Ied 2nd Semester Final Exam Answers Compatibility with Devices
  - Ied 2nd Semester Final Exam Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Ied 2nd Semester Final Exam Answers
  - Highlighting and Note-Taking Ied 2nd Semester Final Exam Answers
  - Interactive Elements Ied 2nd Semester Final Exam Answers
8. Staying Engaged with Ied 2nd Semester Final Exam Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Ied 2nd Semester Final Exam Answers
9. Balancing eBooks and Physical Books Ied 2nd Semester Final Exam Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Ied 2nd Semester Final Exam Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Ied 2nd Semester Final Exam Answers
  - Setting Reading Goals Ied 2nd Semester Final Exam Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ied 2nd Semester Final Exam Answers
  - Fact-Checking eBook Content of Ied 2nd Semester Final Exam Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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