

Chemistry Midterm Study Guide

Periodic Table of the Elements

Legend:

- Alkali Metals
- Alkaline Earth Metals
- Transition Metals
- Noble Gases
- Lanthanides
- Actinides
- Inner Transition Metals

Electron Shells: 1, 2, 3, 4, 5, 6, 7

Electron Configuration: 1s, 2s, 2p, 3s, 3p, 4s, 3d, 4p, 5s, 4d, 5p, 6s, 4f, 5d, 6p, 7s, 5f, 6d, 7p

Electron Configuration: 1s, 2s, 2p, 3s, 3p, 4s, 3d, 4p, 5s, 4d, 5p, 6s, 4f, 5d, 6p, 7s, 5f, 6d, 7p

- Period** - is the horizontal row of elements in the periodic table
ex - Chlorine has an atomic number of 17, so it is in the 3rd period.
- Group** - is the vertical row of elements in the periodic table
ex - Fluorine has an atomic number of 9, so it is in the 17th group.
- Electron Configuration** - is the arrangement of electrons in an atom, represented by a series of numbers and letters (e.g., 1s² 2s² 2p⁶ 3s² 3p⁴)
- Electron Configuration** - The ability to predict the number of electrons in an atom, based on the atomic number, is called electron configuration.

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J Elliott



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