

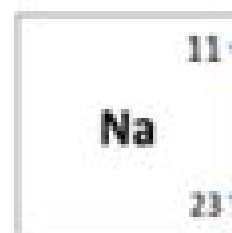
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Atomic number = Number of electrons
= Number of protons

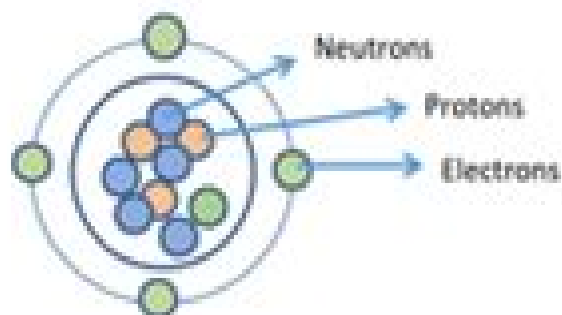
Mass number = Number of neutrons +
Number of protons

To calculate the **number of neutrons** =
Mass number – Atomic number

Top Tip – atomic number is always the **smallest number** but is **NOT** always at the top of the square!

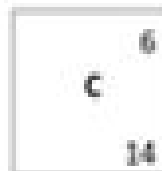
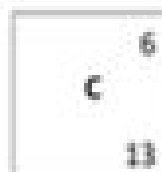
Sub-atomic particles

Particle	Charge	Mass
Proton	+1	1
Neutron	0	1
Electron	-1	1/1836 OR very small/negligible



Top Tip – If there is **more of one type of subatomic particle** in the nucleus, these must be the **neutrons**! Proton number **MUST** equal electron number.

Isotopes



Similarities	Differences
Same atomic number	Different Mass number
Same number of protons	Different number of neutrons
Same number of electrons	Some isotopes are radioactive
Same chemical properties	

Top Tip – The relative atomic mass of an element = the **average value** of all the element's isotopes!

Gcse Additional Chemistry C2 Revision Let

Tom Whipple, James Davies



Gcse Additional Chemistry C2 Revision Let:

This book delves into Gcse Additional Chemistry C2 Revision Let. Gcse Additional Chemistry C2 Revision Let is a vital topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Gcse Additional Chemistry C2 Revision Let, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Gcse Additional Chemistry C2 Revision Let
- Chapter 2: Essential Elements of Gcse Additional Chemistry C2 Revision Let
- Chapter 3: Gcse Additional Chemistry C2 Revision Let in Everyday Life
- Chapter 4: Gcse Additional Chemistry C2 Revision Let in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Gcse Additional Chemistry C2 Revision Let. This chapter will explore what Gcse Additional Chemistry C2 Revision Let is, why Gcse Additional Chemistry C2 Revision Let is vital, and how to effectively learn about Gcse Additional Chemistry C2 Revision Let.
3. In chapter 2, the author will delve into the foundational concepts of Gcse Additional Chemistry C2 Revision Let. This chapter will elucidate the essential principles that need to be understood to grasp Gcse Additional Chemistry C2 Revision Let in its entirety.
4. In chapter 3, this book will examine the practical applications of Gcse Additional Chemistry C2 Revision Let in daily life. This chapter will showcase real-world examples of how Gcse Additional Chemistry C2 Revision Let can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Gcse Additional Chemistry C2 Revision Let in specific contexts. This chapter will explore how Gcse Additional Chemistry C2 Revision Let is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Gcse Additional Chemistry C2 Revision Let. The final chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Gcse Additional Chemistry C2 Revision Let.

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