PART A - CHEMISTRY

- An unknown alcohol is treated with the "Lucas reagent" to determine whether the alcohol is primary, secondary or tertiary. Which alcohol reacts fastest and by what mechanism:
 - (1) tertiary alcohol by S_N1
 - (2) secondary alcohol by S_N2
 - (3) tertiary alcohol by Sx2
 - (4) secondary alcohol by S_N1
- The first ionisation potential of Na is 5.1 eV. The value of electron gain enthalpy of Na + will be:
 - (1) -5.1 eV
 - (2) -10.2 eV
 - (3) + 2.55 eV
 - (4) -2.55 eV
- Stability of the species Li₂, Li₂ and Li₂⁺ increases in the order of:
 - (1) $\text{Li}_{2}^{-} < \text{Li}_{2}^{+} < \text{Li}_{2}$
 - (2) Li₂< Li₂ < Li₂⁺
 - (3) $\text{Li}_{2}^{-} < \text{Li}_{2} < \text{Li}_{2}^{+}$
 - (4) $\text{Li}_2 < \text{Li}_2^+ < \text{Li}_2^-$
- The molarity of a solution obtained by mixing 750 mL of 0.5(M)HCl with 250 mL of 2(M)HCl will be:
 - (1) 1.00 M
 - (2) 1.75 M
 - (3) 0.975 M
 - (4) 0.875 M

- 5. Which of the following is the wrong statement?
 - O₃ molecule is bent.
 - (2) Ozone is violet-black in solid state.
 - (3) Ozone is diamagnetic gas.
 - (4) ONCI and ONO are not isoelectronic.
- Four successive members of the first row transition elements are listed below with atomic numbers. Which one of them is expected to have the highest E⁰_{M³⁺/M²⁺}

value?

- (1) Mn(Z = 25)
- (2) Fe(Z = 26)
- (3) Co(Z = 27)
- (4) Cr(Z = 24)
- A solution of (-)-1-chloro-1phenylethane in toluene racemises slowly in the presence of a small amount of SbCl₅, due to the formation of:
 - (1) carbene
 - (2) carbocation
 - (3) free radical
 - (4) carbanion
- 8. The coagulating power of electrolytes having ions Na⁺, Al³⁺ and Ba²⁺ for arsenic sulphide sol increases in the order:
 - (1) Na + < Ba2+ < Al3+
 - (2) Ba²⁺ < Na⁺ < Al³⁺
 - (3) Al3+ < Na+ < Ba2+
 - (4) $AI^{3+} < Ba^{2+} < Na^{+}$

Engineering Chemistry Paper Solution 2013

Purnima Sharma

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