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(8b)	T	ve:	neactio	on the	at tal	kies p	dace	in	Reactor	2	isa

TiCl, + 4Na → 4NaCl + Ti

Identify the role of sodium in this reaction.

- (1)
- A It catalyses the reaction.
- B It neutralises the titanium(IV) chloride.
- C It reduces the titanium(IV) chloride.
- D It stops reactions with gases in the air.
- (c) Sodium chloride is produced in Reactor 2.

Electrolysis of sodium chloride produces chlorine.

(i) Describe how the electrolysis of sodium chloride produces chlorine.

(4)

Mark Scheme Stuesday 17 January 2012

David Kirk

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