

Unit 4, Lab 2: Boyle's Law

Name: _____

Date: _____

Boyle's Law: In a closed system, the relationship between the volume and pressure of a gas is:

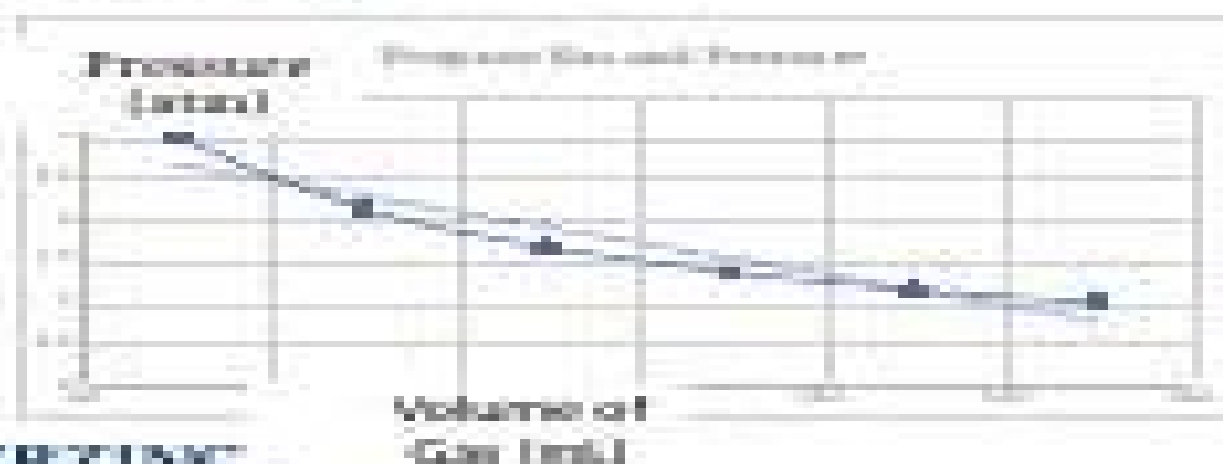
Lab Objectives:

1. Demonstrate the relationship between pressure and volume of a gas.

Pressure (atm)	Volume (mL)
1.00 atm	1.00 mL
0.80 atm	1.25 mL
0.60 atm	1.67 mL
0.40 atm	2.50 mL
0.20 atm	5.00 mL
0.10 atm	10.00 mL

Data Analysis:

1. Plot the pressure (y-axis) versus volume (x-axis) of the gas. The relationship should be a hyperbola. Use the graphing tool below to create the graph and enter the values for the relationship between pressure and volume.



Late Night Lab Answers

Hussin A.Rothana



Late Night Lab Answers:

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1. all real numbers that are less than -3 ... Practice - 3-6 Write a compound inequality that represents each phrase. Graph the solutions.

1. All real numbers that are less than 23 or greater than or equal to 5. Write each set in roster form and in set-builder notation.

Write a compound inequality that represents each phrase. Graph the solutions. 1. all real numbers that are

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that represents the sentence. 1. Four less than a number is greater than ... CompoundIneqA1 03 06 PRG 2.pdf - Name Class

Date ... NameClassDate 3-6 Practice Form G Write a compound inequality that represents each phrase. Graph the solutions.

1. all real numbers that are less than -3 or greater ... 1_6 HW Answers.pdf Aug 20, 2014 — 1-6. Solve each equation. Practice (continued). Absolute Value Equations and Inequalities. Form G. $4-3m=-m-10$. $-2m=-14$. $M=7$. 23. $32x+5=9x-6$. $2x+ ...$

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