32 A scientist investigates two different substances, P and Q.

Substance P completely fills its container but can be compressed.

Substance Q is not in a container but has a definite shape.

In which state is each substance?

	substance P	substance Q
A.	gas	liquid
В	gas	solid
C	liquid	gas
D	liquid	solid

33 A liquid evaporates when molecules leave its surface.

Which molecules leave the surface, and what happens to the temperature of the remaining liquid?

- A The more energetic molecules leave and the temperature falls.
- B The more energetic molecules leave and the temperature rises.
- C The less energetic molecules leave and the temperature falls.
- D The less energetic molecules leave and the temperature rises.
- 34 How is heat transferred in solids?
 - A Heated molecules become less dense and travel to colder areas.
 - B Heated molecules become more dense and travel to colder areas.
 - C Heated molecules vibrate more quickly and cause neighbouring molecules to vibrate more quickly.
 - D Heated molecules vibrate more slowly and cause neighbouring molecules to vibrate more quickly.

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C. C. Yung

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- 1. Understanding the eBook Igcse Combined Science Multiple Choice Question Paper
 - The Rise of Digital Reading Igcse Combined Science Multiple Choice Question Paper
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Igcse Combined Science Multiple Choice Question Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Igcse Combined Science Multiple Choice Question Paper
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Igcse Combined Science Multiple Choice Question Paper
 - Personalized Recommendations
 - Igcse Combined Science Multiple Choice Question Paper User Reviews and Ratings
 - Igcse Combined Science Multiple Choice Question Paper and Bestseller Lists
- 5. Accessing Igcse Combined Science Multiple Choice Question Paper Free and Paid eBooks
 - Igcse Combined Science Multiple Choice Question Paper Public Domain eBooks
 - Igcse Combined Science Multiple Choice Question Paper eBook Subscription Services
 - Igcse Combined Science Multiple Choice Question Paper Budget-Friendly Options
- 6. Navigating Igcse Combined Science Multiple Choice Question Paper eBook Formats

- o ePub, PDF, MOBI, and More
- Igcse Combined Science Multiple Choice Question Paper Compatibility with Devices
- Igcse Combined Science Multiple Choice Question Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of Igcse Combined Science Multiple Choice Question Paper
 - Highlighting and Note-Taking Igcse Combined Science Multiple Choice Question Paper
 - Interactive Elements Igcse Combined Science Multiple Choice Question Paper
- 8. Staying Engaged with Igcse Combined Science Multiple Choice Question Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Igcse Combined Science Multiple Choice Question Paper
- 9. Balancing eBooks and Physical Books Igcse Combined Science Multiple Choice Question Paper
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Igcse Combined Science Multiple Choice Question Paper
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Igcse Combined Science Multiple Choice Question Paper
 - Setting Reading Goals Igcse Combined Science Multiple Choice Question Paper
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Igcse Combined Science Multiple Choice Question Paper
 - Fact-Checking eBook Content of Igcse Combined Science Multiple Choice Question Paper
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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