

PHYSICAL SCIENCE MATTER REVIEW

Identify the letter of the choice that best completes the statement or answers the question!

1. Matter is defined as anything that
 - a. can be seen and touched.
 - b. has mass and takes up space.
 - c. can be weighed.
 - d. contains kinetic or potential energy.
2. A substance that cannot be broken down into simpler substances is
 - a. a compound.
 - b. a mixture.
 - c. an element.
 - d. an atom.
3. A material that can be represented by a chemical formula is
 - a. an element.
 - b. a mixture.
 - c. a homogeneous solution.
 - d. a pure substance.
4. Knowing the chemical properties of a substance will tell you how the substance
 - a. looks.
 - b. smells.
 - c. can be broken down into atoms.
 - d. reacts with other substances.
5. A liquid changes rapidly into a gas at the liquid's
 - a. boiling point.
 - b. freezing point.
 - c. melting point.
 - d. condensation point.
6. A physical property of gold is its
 - a. density.
 - b. reactivity with powerful acids.
 - c. non-flammability.
 - d. None of the above.
7. Grinding quartz crystals down to produce sand is an example of a
 - a. change of state.
 - b. chemical change.
 - c. chemical reaction.
 - d. physical change.
8. The process of a liquid becoming a solid is called
 - a. condensation.
 - b. freezing.
 - c. evaporation.
 - d. melting.
9. Which state of matter has a definite volume, but not shape?
 - a. plasma.
 - b. gas.
 - c. liquid.
 - d. solid.
10. Which state of matter will hold its shape without a container?
 - a. solid.
 - b. liquid.
 - c. gas.
 - d. plasma.
11. The kinetic theory states that the higher the temperature, the faster the
 - a. particles that make up a substance move.
 - b. bonds between atoms break down.
 - c. molecules of gas rush together.
 - d. lighter particles within a substance clump together.
12. The law of conservation of mass states that mass cannot be
 - a. burned.
 - b. changed in form.
 - c. created or destroyed.
 - d. heated or cooled.

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RJ Alexander



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