

Chapter Test B

Chapter: Chemical Bonding

Match the definition on the left of each statement with the term of the choice that best completes the statement or best answers the question.

1. The number of valence electrons in an atom is determined by the number of _____.
 - a. protons in the nucleus
 - b. neutrons
 - c. electrons
 - d. valence
2. A compound is _____ if it has more than one element.
 - a. element
 - b. atomic or molecular
 - c. heterogeneous
 - d. pure
3. Bond length is the distance between two _____.
 - a. their respective potential energy
 - b. their respective atomic weights
 - c. their respective potential energy
 - d. molecules that contain a triple covalent bond
4. Bond energy is the amount of energy that must be added to _____.
 - a. split a molecule into its constituent atoms
 - b. combine atoms
 - c. the number of valence electrons that combine to form a bond
 - d. the number of atoms in a molecule
5. Bond energy is the amount of energy that must be added to _____.
 - a. split a molecule
 - b. split an atom
 - c. combine atoms
 - d. combine a molecule
6. The energy that must be added to break a compound into its constituent atoms is the _____.
 - a. bond energy
 - b. potential energy
 - c. ionization energy
 - d. electron energy

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RD Boyd



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