

Name : _____

Electron Configuration Worksheet

1. Name the elements that have the following electron configurations.

i. $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^5$ _____

ii. $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^4$ _____

2. Write the ground state electron configuration for a neutral atom of the following elements.

i. Oxygen _____

ii. Krypton _____

iii. Chromium _____

3. Using the long method, give the electron configuration of

i. Magnesium (Mg) _____

ii. Potassium (K) _____

iii. Lithium (Li) _____

iv. Nickel (Ni) _____

v. Sulfur (S) _____

4. Identify the following elements.

i. $1s^2 2s^2 2p^2$ _____

ii. $1s^2 2s^2 2p^6$ _____

iii. $[\text{Ar}] 4s^2 3d^{10} 4p^4$ _____

iv. $[\text{Kr}] 5s^2 4d^1$ _____

v. $[\text{Ne}] 3s^2 3p^3$ _____

5. Determine the electron configuration using the short method.

i. Strontium _____

ii. Bromine _____

iii. Zirconium _____

iv. Molybdenum _____

v. Silver _____

Electronic Configuration Worksheet 75 Questions And Answers

Margaret Driscoll



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web solid state physics is the study of rigid matter or solids through methods such as solid state chemistry quantum mechanics crystallography electromagnetism and metallurgy it is the largest branch of condensed matter physics solid state physics studies how the large scale properties of solid materials result from their atomic scale

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matter the stuff that makes up everything

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web figure 3 1 1 3 1 1 matter is usually classified into three classical states with plasma sometimes added as a fourth state from left to right quartz solid water liquid nitrogen dioxide gas the state a given substance exhibits is also a physical property

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