

AICE Marine Science Semester 1 Review

Unit 1 – Water

Key Vocabulary:

- | | | |
|---------------------------|--------------------------|------------------------|
| • Kinetic particle theory | • Polar | • Thermocline |
| • Water vapour | • Solvent | • PH |
| • Proton | • Density | • Acidic |
| • Neutron | • Specific heat capacity | • Alkaline |
| • Electron | • Solute | • Neutral |
| • Nucleus | • Solution | • PH scale |
| • Shells | • Solubility | • Turbulence |
| • Atomic number | • Dissolution | • Dissolved oxygen |
| • Bond | • Salinity | • Phytoplankton |
| • Emergent properties | • Precipitation | • Oxygen minimum layer |
| • Covalent bond | • Run-off | • Water column |
| • Ionic bond | • Evaporation | • Gradient |
| • Hydrogen bond | • Hypersaline | • Thermal insulator |
| • Ion | • Halocline | • Pycnocline |

1. List the 5 oceans and know their locations.
2. Identify the 4 main layers/zones of the ocean and which ones are photic?
3. Identify when you would use the different words: pelagic and benthic?
4. Explain the kinetic particle theory of WATER and the differences between solids, liquids and gases.
5. Identify the following items found in the Ocean
 - a. Key elements
 - b. Key compounds (ionic and covalent)
 - c. Key mixtures
6. Draw the structure of the following atoms (include nucleus, protons, neutrons, and electrons (show arrangement of electrons and shells/rings))
 - a. Oxygen
 - b. Hydrogen
 - c. Carbon
 - d. Sodium
 - e. Chlorine-35
7. Using a diagram describe the covalent bonding in a water molecule including the sharing of electron pairs.

Marine Science Second Semester Exam Review Answers

Karin Nielsen-Saines



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Table of Contents Marine Science Second Semester Exam Review Answers

1. Understanding the eBook Marine Science Second Semester Exam Review Answers
 - The Rise of Digital Reading Marine Science Second Semester Exam Review Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Marine Science Second Semester Exam Review Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Marine Science Second Semester Exam Review Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Marine Science Second Semester Exam Review Answers
 - Personalized Recommendations
 - Marine Science Second Semester Exam Review Answers User Reviews and Ratings
 - Marine Science Second Semester Exam Review Answers and Bestseller Lists
5. Accessing Marine Science Second Semester Exam Review Answers Free and Paid eBooks
 - Marine Science Second Semester Exam Review Answers Public Domain eBooks
 - Marine Science Second Semester Exam Review Answers eBook Subscription Services

- Marine Science Second Semester Exam Review Answers Budget-Friendly Options
- 6. Navigating Marine Science Second Semester Exam Review Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Marine Science Second Semester Exam Review Answers Compatibility with Devices
 - Marine Science Second Semester Exam Review Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Marine Science Second Semester Exam Review Answers
 - Highlighting and Note-Taking Marine Science Second Semester Exam Review Answers
 - Interactive Elements Marine Science Second Semester Exam Review Answers
- 8. Staying Engaged with Marine Science Second Semester Exam Review Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Marine Science Second Semester Exam Review Answers
- 9. Balancing eBooks and Physical Books Marine Science Second Semester Exam Review Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Marine Science Second Semester Exam Review Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Marine Science Second Semester Exam Review Answers
 - Setting Reading Goals Marine Science Second Semester Exam Review Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Marine Science Second Semester Exam Review Answers
 - Fact-Checking eBook Content of Marine Science Second Semester Exam Review Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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