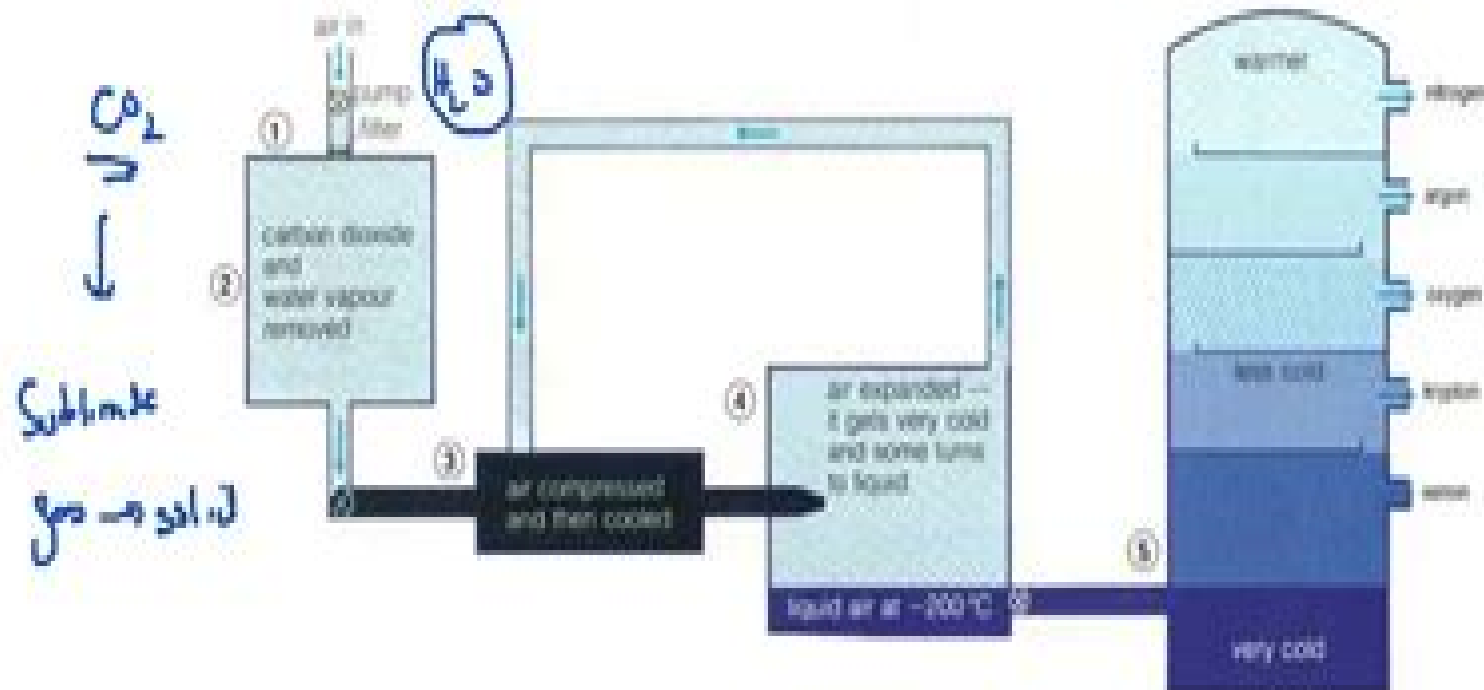


(Extended only) Describe the separation of oxygen and nitrogen from liquid air by fractional distillation

- Fractional distillation of liquid air separates oxygen and nitrogen
- oxygen has a boiling point higher than that of nitrogen
- air is liquefied (around -200°C)
- fractional distillation column temperature is increased until nitrogen boils and rises to the top as a gas (around -196°C)
- the nitrogen is collected and removed
- the temperature is increased until oxygen boils and rises to the top as a gas (around -183°C) and oxygen is collected



Fractional Distillation Of Air Gcse Past Paper

CO Houle



Fractional Distillation Of Air Gcse Past Paper:

This book delves into Fractional Distillation Of Air Gcse Past Paper. Fractional Distillation Of Air Gcse Past Paper is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Fractional Distillation Of Air Gcse Past Paper, encompassing both the fundamentals and more intricate discussions.

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- Chapter 3: Fractional Distillation Of Air Gcse Past Paper in Everyday Life
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- Chapter 5: Conclusion

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