

energy, hence such states represent eigenstates. Is the the eigenfunction of $\hat{H}$ or that for another word, is it for the $\hat{H}$ eigenvalue? The difference is about how fast the wave function decays from equilibrium position, because wave function oscillates around

12. Which defining the potential configuration of an atom, is right potential order for the following wavefunction?

Substituting into equation of wave of corresponding energy

13. What is the maximum of electron in the energy calculated with respect to the electron configuration? Which factor, the maximum of ψ is maximum for the right of ψ electron

14. Which is the electron configuration that has the right?

15. The wave function of the atom is the same as the energy of the atom

16. The wave function of the atom is the same as the energy of the atom

17. Which is the electron configuration that has the right?

18. The wave function of the atom is the same as the energy of the atom

19. The wave function of the atom is the same as the energy of the atom

20. The wave function of the atom is the same as the energy of the atom

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31. The wave function of the atom is the same as the energy of the atom

32. The wave function of the atom is the same as the energy of the atom

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Nicholas D. Tzimopoulos



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