# 5

## **ELECTRONS IN ATOMS**

#### **Practice Problems**

In your notebook, solve the following problems.

#### SECTION 5.1 MODELS OF THE ATOM

- 1. How many sublevets are in the following principal energy levets?
  - $m_e \sim m 1$

 $\mathbf{c}_{2} \cdot \mathbf{z}_{1} = 0$ 

 $\mathbf{b}_{n-1}\mathbf{a}_{1}=2$ 

- $\mathbf{d}_{i} \cdot \mathbf{n}_{i} = 4$
- -----
- 2. How many orbitals are in the following sublevels?
  - a. In subtevel
     b. So subtevel

- d. 4/ sublevet
- es. 7.s subderved

e. did subtlevel.

f. Bay subtlevel

- g. fifth principal energy level
- h. 6d sublevel

 $a_{r}$  ,  $a_{r} = 5$ 

L = 6

- What are the types of sublevels and number of orbitals in the following exemp levels?
  - me or so a

 $c_{ij} = a_i = 3$ 

Section 1975

b. n = 2

 $\mathbf{d}_{i}\cdot m=4$ 

#### SECTION 5.2 ELECTRON ARRANGEMENT IN ATOMS

- 1. Write a complete electron configuration of each atom.
  - a. byrdrogeni

d. barium

g. krypton.

Do. Vonctood burns

ec. becoming

the appearaic

c. magazinesium

L staffbar

L sadon

## SECTION 5.3 PHYSICS AND THE QUANTUM MECHANICAL MODEL

- What is the wavelength of the radiation whose frequency is 5.00 × 10<sup>45</sup> s<sup>-1</sup>? In what region of the electromagnetic spectrum is this radiation?
- An inexpensive laser that is available to the public emits light that has a wavelength of 670 nm. What are the color and frequency of the radiation?
- What is the energy of a photon whose frequency is 2.22 × 10<sup>14</sup> s<sup>-1</sup>?
- What is the frequency of a photon whose energy is 6.00 × 10<sup>-15</sup> F
- Arrange the following types of electromagnetic radiation in order of increasing frequency.
  - a. indrared

e. veisitate Lietar

e. microssoves

b. cosmic rays

d. madio wayses

- ff. admirable to the same of t
- Suppose that your favorite AM radio station broadcasts at a frequency of 1900 kHz. What is the wavelength in meters of the radiation from the station?

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**George Graybill** 

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- 1. Understanding the eBook Electrons And Atoms Worksheet Pearson
  - The Rise of Digital Reading Electrons And Atoms Worksheet Pearson
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Electrons And Atoms Worksheet Pearson
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Electrons And Atoms Worksheet Pearson
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Electrons And Atoms Worksheet Pearson
  - Personalized Recommendations
  - Electrons And Atoms Worksheet Pearson User Reviews and Ratings
  - Electrons And Atoms Worksheet Pearson and Bestseller Lists
- 5. Accessing Electrons And Atoms Worksheet Pearson Free and Paid eBooks
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  - Electrons And Atoms Worksheet Pearson eBook Subscription Services
  - Electrons And Atoms Worksheet Pearson Budget-Friendly Options

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  - Electrons And Atoms Worksheet Pearson Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Electrons And Atoms Worksheet Pearson
  - Highlighting and Note-Taking Electrons And Atoms Worksheet Pearson
  - Interactive Elements Electrons And Atoms Worksheet Pearson
- 8. Staying Engaged with Electrons And Atoms Worksheet Pearson
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Electrons And Atoms Worksheet Pearson
- 9. Balancing eBooks and Physical Books Electrons And Atoms Worksheet Pearson
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Electrons And Atoms Worksheet Pearson
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Electrons And Atoms Worksheet Pearson
  - Setting Reading Goals Electrons And Atoms Worksheet Pearson
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Electrons And Atoms Worksheet Pearson
  - Fact-Checking eBook Content of Electrons And Atoms Worksheet Pearson
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements

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