

## Electron Configuration Crossword

**Directions:** Each entry in the puzzle has an electron configuration as a clue. Write the name of each element with one letter on each line.

[illegible]

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1.  $1s^2 2s^2 2p^6 3s^2 3p^4 4s^2 3d^1$
2.  $1s^2 2s^2 2p^6 3s^2 3p^3$
3.  $[\text{He}] 2s^2 2p^4$
4.  $1s^2 2s^2 2p^6 3s^2 3p^4 4s^2 3d^1$
5.  $[\text{Xe}] 6s^2 4f^{14} 5d^{10}$
6.  $1s^2 2s^2 2p^6 3s^2 3p^4 4s^2 3d^1$
7.  $1s^2 2s^2 2p^6 3s^2$
8.  $1s^2 2s^2 2p^6 3s^2 3p^4 4s^2 3d^{10} 4p^4$
9.  $1s^2 2s^2 2p^6 3s^2 3p^4 4s^2$
10.  $1s^2 2s^2 2p^6 3s^2 3p^4$
11.  $[\text{Kr}] 5s^2 4d^{10}$
12.  $[\text{Rn}] 7s^2 5f^4 6d^1$

10

1.  $[\text{Ne}]3s^2 3p^4 4s^2$
2.  $[\text{Kr}]5s^2 4d^{10} 5p^3$
3.  $1s^2 2s^2 2p^3$
4.  $[\text{Ar}]4s^2 3d^{10}$
5.  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^9$
6.  $1s^2 2s^2 2p^6 3s^2 3p^6$
7.  $1s^2 2s^2 2p^6 3s^2 3p^2$
8.  $[\text{Xe}]6s^2 4f^{14} 5d^9$

# Electron Arrangement Crossword Modern Chemistry

**Padhraic Smyth**



## **Electron Arrangement Crossword Modern Chemistry:**

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