# Q. 1. Select and write the correct answer for the following multiple choice type of questions :

[10]

The product obtained in the following reaction

- $CH_3 CH = CH CH_2 CHO \xrightarrow{H_2/Ni}$ ? is, (a) Pent-3-en-1-ol (b) Pentan-1-ol

(c) Pentan-2-ol

- (d) Pentanal
- (ii) Amongst the following, the solubility of which ionic solid 030 decreases with increase in temperature?

(b) NaBr

(c) Na,SO,

- (d) KCl
- (iii) The correct IUPAC name of Na<sub>3</sub> [AlF<sub>6</sub>] is

- (a) Sodium hexafluoroaluminate (III)
  - (b) Sodium hexafluoroaluminate (II) (c) Sodium hexafluoroaluminium (III)

  - (d) Sodium hexafluoroaluminium (II)
  - (iv) Which of the following acids has highest pKa value?
- (a) Mono chloroacetic acid (b) Dichloroacetic acid (c) Trichloroacetic acid (d) Acetic acid
  - - (v) Number of carbon atoms present in isoprene unit is \_\_\_\_. (b) 5

- (d) 3
- (vi) The colourless transition metal ion amongst the following is

(c) Ni<sup>++</sup>

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**Tobias Bleicker** 

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### **Table of Contents Hsc Unregular Chemistry Question Paper 2bd**

- 1. Understanding the eBook Hsc Unregular Chemistry Question Paper 2bd
  - o The Rise of Digital Reading Hsc Unregular Chemistry Question Paper 2bd
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Hsc Unregular Chemistry Question Paper 2bd
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Hsc Unregular Chemistry Question Paper 2bd
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Hsc Unregular Chemistry Question Paper 2bd
  - Personalized Recommendations
  - Hsc Unregular Chemistry Question Paper 2bd User Reviews and Ratings
  - Hsc Unregular Chemistry Question Paper 2bd and Bestseller Lists
- 5. Accessing Hsc Unregular Chemistry Question Paper 2bd Free and Paid eBooks
  - Hsc Unregular Chemistry Question Paper 2bd Public Domain eBooks
  - Hsc Unregular Chemistry Question Paper 2bd eBook Subscription Services
  - Hsc Unregular Chemistry Question Paper 2bd Budget-Friendly Options
- 6. Navigating Hsc Unregular Chemistry Question Paper 2bd eBook Formats

- ∘ ePub, PDF, MOBI, and More
- Hsc Unregular Chemistry Question Paper 2bd Compatibility with Devices
- Hsc Unregular Chemistry Question Paper 2bd Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Hsc Unregular Chemistry Question Paper 2bd
  - Highlighting and Note-Taking Hsc Unregular Chemistry Question Paper 2bd
  - Interactive Elements Hsc Unregular Chemistry Question Paper 2bd
- 8. Staying Engaged with Hsc Unregular Chemistry Question Paper 2bd
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Hsc Unregular Chemistry Question Paper 2bd
- 9. Balancing eBooks and Physical Books Hsc Unregular Chemistry Question Paper 2bd
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Hsc Unregular Chemistry Question Paper 2bd
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Hsc Unregular Chemistry Question Paper 2bd
  - Setting Reading Goals Hsc Unregular Chemistry Question Paper 2bd
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Hsc Unregular Chemistry Question Paper 2bd
  - Fact-Checking eBook Content of Hsc Unregular Chemistry Question Paper 2bd
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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