

HOLT MODERN CHEMISTRY -- CHAPTER 2 REVIEW

*The following pages contain the bulk (but not all) of the information for the chapter 2 test. Focus on this content, but make sure to review class notes, activities, handouts, questions, etc. If you study this document and NOTHING else, you should at least be able to PASS the test. ***** Test items will be recall, examples, and/or application of this content. ******

OUTCOMES

- 2.1: Describe the scientific method process. *(T & R)*
- 2.2: Measure length, volume, mass, and time using metric units. ($1 \text{ mL} = 1 \text{ cm}^3$ & $1 \text{ L} = 1 \text{ dm}^3$)
(T & R)
- 2.2: Use volume and mass to determine density; compare relative densities of solids, liquids, gases; compare relative densities of pure liquids and solutions. *(F & PK)*
- 2.2: Use density to identify substances in activities involving collection of data and precision and accuracy in measurements. *(F & PK)*
- 2.3: Distinguish between precision and accuracy. *(T & R)*
- 2.3: Determine and perform calculations using significant figures. *(F & PK)*
- 2.3: Convert from numbers to scientific notation and reverse. *(F & PK)*

2.1 SCIENTIFIC METHOD

- **Vocabulary**
 - **scientific method** -- a series of steps followed to solve problems, including collecting data, formulating a hypothesis, testing the hypothesis, and stating conclusions
 - **system** -- a set of particles or interacting components considered to be a distinct physical entity for the purpose of study
 - **hypothesis** -- an explanation that is based on prior scientific research or observations and that can be tested
 - **model** -- a pattern, plan, representation, or description designed to show the structure or workings of an object, system, or concept
 - **theory** -- an explanation for some phenomenon that is based on observation, experimentation, and reasoning
- **Chapter Highlights**
 - The scientific method is a logical approach to solving problems that lend themselves to investigation.
 - A hypothesis is a testable statement that serves as the basis for predictions and further experiments. It is usually written as an if/then statement.
 - A theory is a broad generalization that explains a body of known facts or phenomena.
 - Communication is a very important aspect of all parts of the scientific method, including successful and/or unsuccessful results.

2.2 UNITS OF MEASUREMENT

- **Vocabulary**
 - **quantity** -- something that has magnitude, size, or amount
 - **SI** -- Le Systeme International d'Unites, or the International System of Units, which is the measurement system that is accepted worldwide
 - **weight** -- a measure of the gravitational force exerted on an object; its value can change with the location of the object in the universe
 - **derived unit** -- a unit of measure that is a combination of other measurements
 - **volume** -- a measure of the size of a body or region in three-dimensional space
 - **density** -- the ratio of the mass of a substance to the volume of the substance; often expressed as grams per cubic centimeter for solids and liquids and as grams per liter for gases
 - **conversion factor** -- a ratio that is derived from the equality of two different units and that can be used to convert from one unit to the other
 - **dimensional analysis** -- a mathematical technique for studying dimensions of physical quantities

Holt Modern Chemistry Test B Answers

Robert R. Redfield



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the reaction of an acid and a base is called a neutralization reaction although acids and bases have their own unique

chemistries the acid and base cancel each other's chemistry to produce a rather innocuous substance water in fact the general reaction between an acid and a base is

$$\text{acid} + \text{base} \rightarrow \text{water} + \text{salt}$$

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