

# Practice A

For use with pages 16–18

Draw a sketch and label as needed.

- Three collinear points, A, B, and C.
- Coplanar points W, X, Y, and Z.



- $\vec{MN}$  intersecting  $\vec{PQ}$  at point R.
- Opposite rays,  $\vec{JR}$  and  $\vec{JC}$ .



Decide whether the statement is true or false.

- Point X lies on line m.
- Point W lies on line m.
- Point V lies on line l.
- Point Y lies on line l.
- X, Y, and Z are collinear.
- X, Y, and Z are coplanar.
- V, Y, and X are collinear.
- V, Y, and X are coplanar.

T  
F  
T  
T

F  
F  
F  
F



Name a point that is collinear with the given points.

- B and E
- D and G
- H and E
- A and D
- F and H
- A and C
- G and C
- B and C

H  
A  
B  
G

G  
E  
D  
A



Name a point that is coplanar with the given points.

- M, N, and O
- M, N, and R
- T, Q, and M
- T, Q, and R
- T, S, and R
- T, S, and O
- O, S, and R
- O, P, and M

P  
Q  
N  
O

Q  
S  
P  
N



In Exercises 29–34, complete the sentence.

- Collinear points are points that \_\_\_\_\_. *lie on the same line*
- Coplanar points are points that \_\_\_\_\_. *lie on the same plane*
- $\overline{XY}$  consists of the endpoints X and Y and all points on the line  $\overleftrightarrow{XY}$  that lie \_\_\_\_\_. *between them*
- $\vec{MN}$  consists of the initial point M and all points on the line  $\overleftrightarrow{MN}$  that lie \_\_\_\_\_. *between M and N*
- Two rays or segments are collinear if they \_\_\_\_\_. *lie on the same line*
- $\vec{PQ}$  and  $\vec{PR}$  are opposite rays if \_\_\_\_\_. *same endpoint P and opposite direction*
- Explain the difference between  $\vec{BC}$  and  $\overleftrightarrow{BC}$ .

$\vec{BC}$  = endpoint B, endpoint C  
 $\overleftrightarrow{BC}$  = line

# Holt Geometry Practice 6

**Marco Casella**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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web reaction rates google classroom  $\text{mg} + \text{s} \rightarrow 2 \text{hcl} + \text{a} + \text{q}$   $\text{mgcl}_2 + \text{a} + \text{q} + \text{h}_2 + \text{g}$  the reaction between  $\text{mg} + \text{s}$  and  $\text{hcl} + \text{a} + \text{q}$  is represented by the equation above in a kinetics experiment a 0.080 g sample of  $\text{mg} + \text{s}$  ribbon is added to an erlenmeyer flask containing 50 ml of 3.0 M  $\text{hcl} + \text{a} + \text{q}$  at 25 °C

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