

# Thermodynamics

## Section Quiz: The First Law of Thermodynamics

Write the letter of the correct answer in the space provided.

- \_\_\_\_\_ 1. Which concept does the first law of thermodynamics describe?
  - a. conservation of mass
  - b. conservation of energy
  - c. energy being transformed
  - d. conservation of momentum
- \_\_\_\_\_ 2. Which concept allows  $Q = 0$  for the first law of thermodynamics when the heat of  $Q_1 = -Q_2$ ?
  - a. an isolated system processes
  - b. an adiabatic process
  - c. an isobaric process
  - d. an isochoric process
- \_\_\_\_\_ 3. Which concept allows  $W = 0$  for the first law of thermodynamics when the heat of  $Q_1 = Q_2$ ?
  - a. an isobaric process
  - b. an isochoric process
  - c. an isobaric process
  - d. an isochoric process
- \_\_\_\_\_ 4. Which process is isothermal,  $Q = W = 0$ , so that the first law of thermodynamics takes the form of  $Q_1 = Q_2$ ?
  - a. an isobaric process
  - b. an isochoric process
  - c. an isobaric process
  - d. an isochoric process
- \_\_\_\_\_ 5. Which concept allows  $Q_1 = 0$  so that the first law of thermodynamics takes the form of  $Q_2 = W$ ?
  - a. an isobaric process
  - b. an isochoric process
  - c. an isobaric process
  - d. an isochoric process
- \_\_\_\_\_ 6. A process might follow the  $Q_1 = 0$  of energy is heat, giving up  $Q_2 = 0$  of energy as heat, the surroundings, and some  $Q_3 = 0$  of work. What further change is the internal energy of the system?
  - a. zero J
  - b. 1000 J
  - c. 2000 J
  - d. 3000 J

# Holt Physical Science Quiz Section Temperature Answers

**J Ma**



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