

# 2024 CHEM 112- FINAL EXAM ACTUAL QUESTIONS AND ANSWERS WITH COMPLETE SOLUTION

quickly - gases react more [slowly/quickly] than liquids

concentration, temperature, catalyst, diffusion rate - factors affecting reaction rates

reaction rate - the speed of a chemical reaction in molarity/second

change in concentration of B/change in time - average rate of appearance of B

instantaneous rate - the rate of reaction at a particular moment

initial rate - rate at time = 0

rate law - an equation that shows how the rate depends on the concentrations of reactants

$K[A]^m \times [B]^n$  - general form of a rate law

rate constant - the units of this can be found by dividing M/s by the units of concentration

zero - if a reaction is unchanged by concentration, it is this order

first - if the rate change is proportional to concentration change, the reaction is this order

second or greater - if rate change is exponential, the reaction is this order

# Exam Question On Chem112

**Annelies Wilder-Smith**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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