Chapter 2 - Essential Chemistry for Biology

(text) - Important

Some Basic Chemistry

- Matter: Anything that occupies space nd has mass.
 - Three states: Solid, Liquid, and Gas-
 - Compromised of elements -> substances that cannot be broken down into other substance.
 - All living things share these elements n varying proportion:
 - Sulphor
 - 2) Phosphorus
 - Oxygen.
 - Nitrogen.
 - Carbon
 Hydrogen
- 25 essential elements
- Four elements make up about 96% of the weight of most cells:
 - 1) Oscyrgom
 - 2) Carbon
 - Hydrogen
 - 4) Nitrogen
- · Trace Elements: required in very small amounts.

Atoms

 Each element consists of one kind of atom -> the smallest unit of matter that still retains its chemical properties.

Basic Structure of an Atom

- Atoms are composed of three subatomic particles:
 - Protons (Positive Charge) Found in the nucleus
 - Neutrons (Electrically Neutral) Found in the nucleus
 - Electrons (Negative Charge) Found in the Electron Cloud
- Elements are composed of atoms that differ in the number of subatomic particles.
- · An atom is characterized by:
 - Atomic Number: the number of Protons
 - Mass Numbers: Sum of the number of Protons plus neutrons
 - * Proton 1 Dalton
 - Neutron 1 Daliton
 - Electron Negligible

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Brendan G. Carr

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