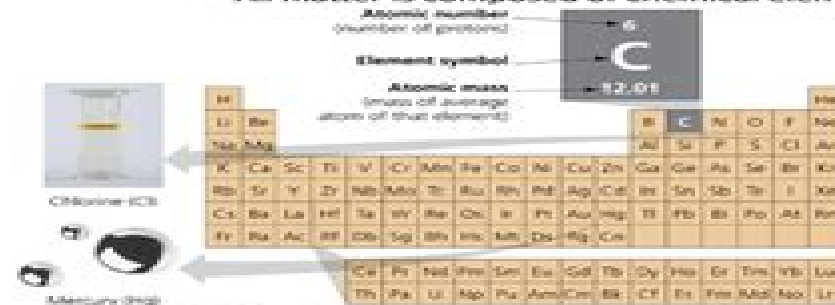
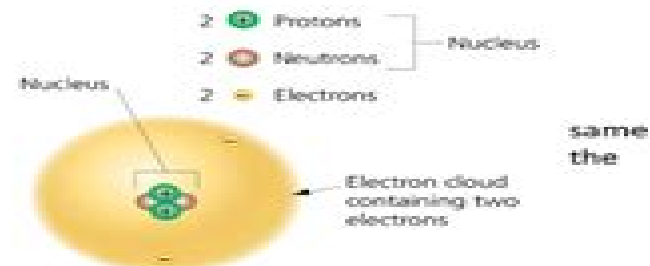


Chapter 2 Bio Essential Chemistry For Biology

- Take any biological system apart, and you eventually end up at the chemical level.
 - Matter** is anything that occupies space and has mass.
 - Mass** is a measure of the amount of material in an object.
 - An **element** is a substance that cannot be broken down into other substances by chemical reactions.
 - All matter is composed of chemical elements.



- Of the 92 naturally occurring elements, 25 are essential to people.
 - Four of these elements make up about 96% of the weight of the body.
 - Trace elements** are required in only very small amounts and are essential for life.
 - Elements can combine to form **compounds**, substances that contain two or more elements in a fixed ratio.
- Each element consists of one kind of atom.
- An **atom** is the smallest unit of matter that still retains the properties of an element.
- Atoms are composed of subatomic particles.
 - A **proton** is positively charged.
 - An **electron** is negatively charged.
 - A **neutron** is electrically neutral.
- All atoms of a particular element have the unique number of protons. This number is element's **atomic number**.
- An atom's **mass number** is the sum of the number of protons and neutrons in its nucleus.
 - Isotopes** have the same number of protons and behave identically in chemical reactions, but they have different numbers of neutrons.
 - A **radioactive isotope** is one in which the nucleus decays spontaneously.
- Only electrons are directly involved in chemical reactions.



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K Morrison



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