

Science 101

Chapter Test A

Teacher Notes and Answers

Subatomic Physics

Questions 1-10 are (Answers)

1. A
 2. A
 3. A
 4. B
 5. B
 6. B
 7. B
 8. B
 9. B
 10. B
- Answers:
1. Mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
2. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
3. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
4. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
5. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
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7. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
8. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
9. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
10. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.

11. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
12. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
13. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
14. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
15. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
16. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
17. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
18. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
19. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.
20. The mass number of a nucleus is the sum of the number of protons and neutrons in the nucleus.

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Lingsheng Yao



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