

# FORD

125 AND 145  
LAWN AND GARDEN  
TRACTORS

## OPERATOR'S MANUAL



FORD TRACTOR OPERATIONS  
FORD MOTOR COMPANY

# Ford Lgt 125 Service Manual

**Minjie Lin**



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web we can balance it for charge by adding two electrons to the right side of the equation so that the net charge on each side is 0 oxidation ni s ni a 2 a q 2 e now that the oxidation half reaction is balanced it tells us that two electrons are produced for every atom of nickel oxidized but where do those electrons go

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web balance each half reaction in basic solution 4 5 6 7  $\text{Cr}_2\text{O}_7^{2-}$  2  $\text{NO}_3^-$  2  $\text{MnO}_2$   $\text{Cr}_3$   $\text{NO}_3$   $\text{SO}_2$   $\text{Mn}_2\text{O}_3$  balance each redox reaction in acid solution using the half reaction method 8  $\text{H}_2\text{O}_2$   $\text{Cr}_2\text{O}_7^{2-}$  2 9  $\text{TeO}_3$  2  $\text{N}_2\text{O}_4$   $\text{O}_2$   $\text{Cr}_3$   $\text{Te}$  10  $\text{ReO}_4$  11  $\text{PbO}_2$   $\text{IO}_2$   $\text{Pb}_2$  12  $\text{As}_2\text{O}_3$   $\text{IO}_3^-$   $\text{Re}$   $\text{IO}_3^-$   $\text{H}_2\text{AsO}_4$

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web in the redox equation below that occurs in our stomach which of the following pairs identifies the reducing and oxidizing agents respectively  $\text{C}_6\text{H}_8\text{O}_6$   $2\text{H}_2\text{NO}_2$   $\text{C}_6\text{H}_6\text{O}_6$   $2\text{H}_2\text{O}_2$   $2\text{NO}$  choose 1 answer  $\text{C}_6\text{H}_8\text{O}_6$  and  $2\text{NO}$  a  $\text{C}_6\text{H}_8\text{O}_6$  and  $2\text{NO}_2$  b  $2\text{H}_2\text{O}_2$  and  $\text{C}_6\text{H}_8\text{O}_6$  c  $2\text{H}_2\text{NO}_2$  and  $\text{C}_6\text{H}_6\text{O}_6$  d  $2\text{H}_2\text{O}_2$  and  $2\text{H}_2\text{NO}_2$

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