


- (c) 350 cm, convergent  
(d) 150 cm, divergent
16. Three resistances A, B and C each of  $5\Omega$  are connected to a battery of 10 V as shown in the figure. Calculate the current through C:
- 
- (a) 2 amp (b) 1 amp  
(c) 3 amp (d) 0.6 amp
17. The device used to generate electrical energy is:
- (a) Electric motor (b) Generator  
(c) Galvanometer (d) Voltmeter
18. The number of molecules in 11 g of  $\text{CO}_2$  are:
- (a)  $0.25 \times 10^{23}$  (b)  $0.50 \times 10^{23}$   
(c)  $1.00 \times 10^{23}$  (d)  $1.51 \times 10^{23}$
19. The value of charge / mass ratio of electron was determined by:
- (a) W.K. Roentgen (b) J.J. Thomson  
(c) Marie Curie (d) Niels Bohr
20. Which of the following reactions is an example of combination reaction?
- (a)  $\text{Zn(s)} + \text{CuSO}_4(\text{aq}) \rightarrow \text{ZnSO}_4(\text{aq}) + \text{Cu(s)}$   
(b)  $\text{BaCl}_2(\text{aq}) + \text{Na}_2\text{SO}_4(\text{aq}) \rightarrow \text{BaSO}_4(\text{s}) + 2\text{NaCl}(\text{aq})$   
(c)  $\text{CaCO}_3(\text{s}) \rightarrow \text{CaO(s)} + \text{CO}_2(\text{g})$   
(d)  $\text{C(s)} + \text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g})$
21. The element 'X' and 'Y' have atomic numbers 12 and 17 respectively. Element 'X' reacts with element 'Y' to form compound with molecular formula:
- (a) XY (b)  $\text{XY}_2$   
(c)  $\text{XY}_3$  (d)  $\text{X}_3\text{Y}_3$
22. The atomic radius (pm) of Li, Na, K and Rb varies in the order:
- (a)  $\text{Na} < \text{K} < \text{Rb} < \text{Li}$  (b)  $\text{K} < \text{Na} < \text{Li} < \text{Rb}$   
(c)  $\text{Li} < \text{Na} < \text{K} < \text{Rb}$  (d)  $\text{Rb} < \text{K} < \text{Na} < \text{Li}$
23. The electron affinity value ( $\text{kJ mol}^{-1}$ ) of fluorine (F) is less than:
- (a) Hydrogen (H) (b) Lithium (Li)  
(c) Oxygen (O) (d) Chlorine (Cl)
24. Which of the following is an example of strong electrolyte?
- (a)  $\text{H}_2\text{CO}_3$  (b)  $\text{NH}_4\text{OH}$   
(c) NaCl (d)  $\text{COOH} - \text{COOH}$
25. If pH of a solution changes from 5 to 4, the change in hydrogen ion concentration shall be:
- (a) two times (b) five times  
(c) ten times (d) twenty times
26. Bleaching powder is manufactured by the reaction of:
- (a)  $\text{CaCl}_2$  and  $\text{CaCO}_3$  (b)  $\text{Cl}_2$  and  $\text{Ca(OH)}_2$
27. (c)  $\text{Cl}_2$  and  $\text{CaSO}_4$  (d)  $\text{Cl}_2$  and  $\text{Ca(HCO}_3)_2$   
The hard glass is obtained by fusing:
- (a) soda ash, sand and limestone  
(b) a mixture of sand, lime, borax and alkali carbonates  
(c) potassium carbonate and limestone  
(d) potassium carbonate, lead oxide and sand
28. Which of the following metals is most reactive?
- (a) Aluminium (b) lead  
(c) Mercury (d) Silver
29. Which of the following alloys contains chromium?
- (a) Steel (b) Stainless steel  
(c) Magnadium (d) Brass
30. Which of the following is monomer of natural rubber?
- (a) Chloroethene (b) Chloroprene  
(c) Isoprene (d) Buta-1, 3-diene
31. Alkaline  $\text{KMnO}_4$  oxidises propanone to:
- (a) propanoic acid (b) ethanoic acid  
(c) methanoic acid (d) oxalic acid
32. Which of the following metals can displace Zn from  $\text{ZnSO}_4$  solution?
- (a) Calcium (b) Copper  
(c) Iron (d) Mercury
33. Heating of sodium ethanoate with soda lime yields:
- (a) ethane (b) methane  
(c) ethanol (d) methanol
34. The compound formed by the reaction of ethyne with bromine is:
- (a)  $\text{Br}-\text{CH}=\text{CH}-\text{Br}$  (b)  $\text{Br}-\text{CH}_2-\text{CHBr}_2$   
(c)  $\text{CH}_2=\text{CH}-\text{Br}$  (d)  $\text{Br}_2\text{CH}-\text{CHBr}_2$
35. Cell were first discovered by Robert Hooke in the year:
- (a) 1665 (b) 1674 (c) 1831 (d) 1839
36. The plant cells have a rigid cell wall that lies:
- (a) Outside the plasma membrane  
(b) Inside the plasma membrane  
(c) In between the plasma membranes  
(d) None of the above
37. During mitosis nucleolus and nuclear membrane are lost in stage:
- (a) Prophase (b) Metaphase  
(c) Anaphase (d) Telophase
38. An example of simple and permanent tissue is:
- (a) Xylem (b) Phloem  
(c) Sclerenchyma (d) All the above
39. The shape of squamous epithelial tissue is:
- (a) Cubical (b) Flattened  
(c) Pillar like (d) None of the above
40. Fresh water sponge belongs to phylum:
- (a) Arthropoda (b) Annelida

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**JS Bruner**



### **Exam Question Paper Grade11 Caps 2013:**

School Science Practical Work in Africa Umesh Ramnarain, 2020-06-01 School Science Practical Work in Africa presents the scope of research and practice of science practical work in African schools. It brings together prominent science educators and researchers from Africa to share their experience and findings on pedagogical innovations and research informed practices on school science practical work. The book highlights trends and patterns in the enactment and role of practical work across African countries. Practical work is regarded as intrinsic to science teaching and learning and the form of practical work that is strongly advocated is inquiry based learning which signals a definite paradigm shift from the traditional teacher dominated to a learner centered approach. The book provides empirical research on approaches to practical work, contextual factors in the enactment of practical work and professional development in teaching practical work. This book will be of great interest to academics, researchers and post graduate students in the fields of science education and educational policy. Proceedings of the 2nd UPY International Conference on Education and Social Science (UPINCESS 2023) Ari Kusuma Wardana, 2023-12-31 This is an open access book. Technology has had a huge impact on education and social dynamics around the world. Since the introduction of information and communication technologies (ICTs) such as the internet, smartphones and social media, much has changed in the way we learn and interact with others. In education, technology has enabled students and teachers to access more diverse and engaging educational resources. With internet access, students and teachers can access educational resources from all over the world and learn about various topics and disciplines. In addition, technology also enables distance learning which allows students and teachers to connect from different locations. However, there are also negative impacts of using technology in education. For example, excessive use of social media and the internet can cause concentration disorders and addiction in students. In addition, too much reliance on technology can also reduce students' ability to think critically and solve problems independently. In social dynamics, technology has allowed us to connect with people from all over the world and paved the way for greater collaboration and cultural exchange. However, technology has also brought negative impacts on social dynamics such as the spread of fake news or hoaxes, cyberbullying and social media addiction. In this regard, technology can be a very useful tool if used wisely and with proper understanding of its impact. Therefore, to support this, Universitas PGRI Yogyakarta was holding an international conference and Call for Papers: The 2nd UPY International Conference on Education and Social Science (UPINCESS) The Impact of Technology on Education and Social Dynamics on September 27, 2023. The conference aims to discuss in depth about the impact of technology on education and social dynamics as well as finding appropriate solutions and strategies to minimize its negative impact and maximize its benefits. Experts, researchers and practitioners in the field of education and technology from around the world can share their knowledge and experience in this conference and can encourage collaboration and innovation in this field. **The Cultivator & Country Gentleman**, 1888 **MacRae's Blue Book**, 1990

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