

## Chapter 3: Matter and Energy – Study Guide

\*Do not forget about the extra resources available to help you study and prepare for this test!

- Textbook pages 81- 93 (answers to odd questions in the back of the book)
- Mastering Chemistry Study Area: <http://www.pearsonmylabandmastering.com/northamerica/>
- Miss Marnik's Website: <http://www.northallegheny.org/Page/16665>
- Labs, problem packets, and homework assignments

### Section 1 – Classifying Matter: (text pages 57-62)

Need to be able to:

- Define matter, atoms, molecules, element, compound, and mixture.
- Classify matter as and distinguish between solid, liquid, or gas.
- Classify matter as element, compound, or mixture.

### Practice Problems:

1. Match the following terms with the phrases below:

- |  |                          |
|--|--------------------------|
| a. State of matter in which atoms or molecules pack close to each other in fixed locations                             | 1. heterogeneous mixture |
| b. State of matter in which atoms or molecules are packed close together but are free to move around and by each other | 2. homogeneous mixture   |
| c. Substance composed of only one type of atom or molecule   | 3. liquid                |
| d. Matter composed of two or more different types of atoms or molecules combined in variable proportions               | 4. mixture               |
| e. Mixture that has two or more regions with different compositions  | 5. pure substance        |
| f. Mixture that has the same composition throughout  | 6. solid                 |

2. Which of the following can be classified as matter?

- |          |            |                     |           |
|----------|------------|---------------------|-----------|
| a. atoms | c. wind    | e. potential energy | g. carbon |
| b. gas   | d. liquids | f. temperature      | h. light  |

3. Classify each type of matter as a pure substance or a mixture. If it is a pure substance, classify it as an element or a compound. If it is a mixture, classify it as homogeneous or heterogeneous.

- |                                  |               |
|----------------------------------|---------------|
| a. limestone ( $\text{CaCO}_3$ ) | e. Chex mix   |
| b. orange juice with pulp        | f. Table salt |
| c. helium                        | g. Brass      |
| d. Pacific ocean                 | h. oxygen     |

# Matter And Energy Study Guide

**Leia Levy**



## **Matter And Energy Study Guide:**

The book delves into Matter And Energy Study Guide. Matter And Energy Study Guide is an essential topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Matter And Energy Study Guide, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Matter And Energy Study Guide
- Chapter 2: Essential Elements of Matter And Energy Study Guide
- Chapter 3: Matter And Energy Study Guide in Everyday Life
- Chapter 4: Matter And Energy Study Guide in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Matter And Energy Study Guide. This chapter will explore what Matter And Energy Study Guide is, why Matter And Energy Study Guide is vital, and how to effectively learn about Matter And Energy Study Guide.

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