Name: Date:

Limiting Reagent Worksheet

Answers

Given the following reaction:

$$C_1H_A + \underline{5} O_1 \rightarrow \underline{3} CO_1 + \underline{4} H_1O_2$$

Balance the reaction.

Suppose 14.8 g of C₃H₈ reacts with 3.44 g of O₂. Determine the limiting reagent.

$$14.8 \text{ g C}_{3}H_{8} \times \frac{1 \text{ mol C}_{3}H_{8}}{44.09 \text{ g C}_{3}H_{8}} \times \frac{4 \text{ mol H}_{2}O}{1 \text{ mol C}_{3}H_{8}} \times \frac{18.02 \text{ g H}_{2}O}{1 \text{ mol H}_{2}O} = 24.2 \text{ g H}_{2}O$$

$$3.44 \text{ g O}_{2} \times \frac{1 \text{ mol O}_{2}}{32 \text{ g O}_{2}} \times \frac{4 \text{ mol H}_{2}O}{5 \text{ mol O}_{2}} \times \frac{18.02 \text{ g H}_{2}O}{1 \text{ mol H}_{2}O} = 1.55 \text{ g H}_{2}O$$

Since O, produces less quantity of H,O than C,H, does, it is the limiting reagent.

(b) Determine the amount of CO, produced.

2 Given the following reaction:

Balance the reaction.

What is the limiting reagent if 2.2 g of Mg reacts with 4.5 L of O, at STP?

2.2 g Mg *
$$\frac{1 \text{ mol Mg}}{24.3 \text{ g Mg}}$$
 * $\frac{2 \text{ mol MgO}}{2 \text{ mol Mg}}$ = 0.0905 mol MgO
4.5 L O₂ * $\frac{1 \text{ mol O}_2}{22.4 \text{ L O}_2}$ * $\frac{2 \text{ mol MgO}}{1 \text{ mol O}_2}$ * = 0.401 mol MgO

Since Mg produces less MgO than O, Mg is the limiting reagent.

Determine the amount of MgO produced in grams.

Limiting Reactant Work Sheet

Katherine E. Hagerman

Limiting Reactant Work Sheet:

General Chemistry Workbook Daniel C. Tofan,2010-07-28 This workbook is a comprehensive collection of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and questions It contains a total of 197 learning objectives grouped in 28 lessons and covering the vast majority of the types of problems that a student will encounter in a typical one year chemistry course It also contains a fully solved 50 question practice test which gives students a good idea of what they might expect on an actual final exam covering the entire material **Working with Chemistry** Donald J. Wink, Sharon Fetzer-Gislason, Julie Ellefson Kuehn, 2004-02-20 With this modular laboratory program students build skills using important chemical concepts and techniques to the point where they are able to design a solution to a scenario drawn from a professional environment The scenarios are drawn from the lives of people who work with chemistry every day ranging from field ecologists to chemical engineers and include many health professionals as well

Holt Chemistry Holt Rinehart & Winston, 2003-01-24

ChemDiscovery Teacher Edition Olga I. Agapova, 2002 Science and Technology Handbook Part 2 Chandan Sengupta, A Workbook for CBSE Students of Class X Chandan Sengupta Title Science and Technology Handbook Part 2 Author Chandan Sengupta Year of Publication 2025 Date of Publication 14 02 2025 Format Book Country INDIA ISBN 978 93 342 2360 6 This book has been published with all reasonable efforts duly taken up to make the material error free after getting the consent of the author No part of this book shall be used reproduced in any manner whatsoever without written permission from the author except in the case of brief quotations embodied in critical articles and reviews What we expect from our fellow student and what are the facilities we provide them should have proper links for ensuring the maximum return of our efforts We even come across instances during which children may revolt repeatedly to avoid scheduled intensive learning programmes duly planned for them For efficient handling of such jobs we should go on planning content delivery mechanism on the basis of student centered curriculum design It will even link up our plan with those of other fellow faculty members for making the effort a vibrant one The work book like this one and others of similar category has a comprehensive plan of addressing content areas duly specified by the boards of studies Answer sheets are there for some selected test papers Rest of the other sheets kept off the side for enabling the exploratory drive of fellow students active We are expecting their active participation in the learning and facilitation drives It is true that this workbook cannot follow the content areas exclusively prescribed for the aspirants of the particular age group The purpose of the incorporations of varying types of activities is to expose the fellow students to some forthcoming challenges It will definitely imply a sort of impression in the mind of the student and enable them to grasp through higher challenges with subtle easiness

Handbook on Material and Energy Balance Calculations in Material Processing, Includes CD-ROM Arthur E. Morris, H. Alan Fine, Gordon Harold Geiger, 2011-09-06 This book approaches the subject of material and energy balances

from two directions First it emphasizes the fundamental principles of the conservation of mass and energy and the consequences of these two principles Second it applies the techniques of computational chemistry to materials processing and introduces new software developed by the author especially for material and heat balances. The third edition reflects the changes in the professional engineer's practice in the last 30 years reflecting the dramatic shift away from metallurgical engineering and the extractive industry towards materials engineering A large and growing number of recent graduates are employed in such fields as semiconductor processing environmental engineering and the production and processing of advanced and exotic materials for aerospace electronic and structural applications. The advance in computing power and software for the desktop computer has significantly changed the way engineers make computations and the biggest change comes from the computational approach used to solve problems The spreadsheet program Excel is used extensively throughout the text as the main computational engine for solving material and energy balance equations and for statistical analysis of data The use of Excel and the introduction of the add in programs enables the study of a range of variables on critical process parameters and emphasis is placed on multi device flowsheets with recycle bypass and purge streams whose material and heat balance equations were previously too complicated to solve by the normally used hand calculator The Excel based program FlowBal helps the user set up material and heat balance equations for processes with multiple streams and A Concrete Stoichiometry Unit for High School Chemistry Jennifer Louise Pakkala, 2006 units Thermodynamics of Phase and Reaction Equilibria Ismail Tosun, 2021-06-17 The Thermodynamics of Phase and Reaction Equilibria Second Edition provides a sound foundation for understanding abstract concepts of phase and reaction equilibria e g partial molar Gibbs energy fugacity and activity and shows how to apply these concepts to solve practical problems using numerous clear examples Available computational software has made it possible for students to tackle realistic and challenging problems from industry The second edition incorporates phase equilibrium problems dealing with nonideal mixtures containing more than two components and chemical reaction equilibrium problems involving multiple reactions Computations are carried out with the help of Mathcad Clear layout coherent and logical organization of the content and presentation suitable for self study Provides analytical equations in dimensionless form for the calculation of changes in internal energy enthalpy and entropy as well as departure functions and fugacity coefficients All chapters have been updated primarily through new examples Includes many well organized problems with answers which are extensions of the examples enabling conceptual understanding for quantitative real problem solving Provides Mathcad worksheets and subroutines Includes a new chapter linking thermodynamics with reaction engineering A complete Instructor's Solutions Manual is available as a textbook resource STOICHIOMETRY NARAYAN CHANGDER, 2024-04-01 Note Anyone can request the PDF version of this practice set workbook by emailing me at cbsenet4u gmail com You can also get full PDF books in quiz format on our youtube channel https www youtube com smartquiziz I will send you a PDF version of this workbook This book has

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