Stoichiometry Limiting Reactant

Answers

How many grams of calcium phosphate can be produced from the reaction between 2.50 L of 0.250 M calcium chloride with an excess of phosphoric acid?

2 How many milliliters of 1.50 M nitric acid is required to react with 100.0 g of cuprous oxide?

(3) Consider the following unbalanced reaction:

Balance the reaction. 3.25 g of NH, is allowed to react with 3.5 g of O2.

Which reactant is the limiting reagent?

O, is the limiting reagent.

b How many grams of NO are formed?

2.63 g of NO are formed.

E How much excess reagent remains after the reaction?

$$3.5 \text{ g O}_2 \times \frac{1 \text{ mol O}_2}{32 \text{ g O}_2} \times \frac{4 \text{ mol NH}_3}{5 \text{ mol O}_2} \times \frac{17.03 \text{ g NH}_3}{1 \text{ mol NH}_3} = 1.49 \text{ g NH}_3 \text{ used}$$

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