Name_____

# Second Semester Final Exam

**Multiple Choice:** Please mark the answer to the multiple choice questions on your bubble sheet.

1) The molar mass of magnesium cyanide, $Mg(CN)_2$, is approximately
   A) 50.3 g
   B) 64.3 g
   C) 76.3 g
   D) 100.6 g
2) The total number of atoms in $(NH_4)_3BO_3$ is
   A) 12
   B) 17
   C) 19
3) What is the mass in grams of 5.90 mol $C_8H_{18}$?
   A) 0.0512 g
   B) 389 g
   C) 673 g
   D) 19.4 g
4) How many atoms are there in 1 mole of helium?
   A) $6.02 \times 10^{23}$ atoms
   B) 22.4 atoms
   C) 4.0 atoms
   D) 0.0821 atoms
5) The percent composition of silver (Ag) in $AgNO_3$ is
   A) 0.625 %
   B) 1.57 %
   C) 62.5 %
   D) 157 %
   E) 169.9 %
6) Which of the following is an empirical formula?
   A) $C_2H_4$
   B) $C_3H_6$
   C) $CH_4$
   D) $C_3H_6$
7) Which of the following could be a molecular formula for the empirical formula $CH_2$
   A) $C_3H_6$
   B) $C_6H_3$
   C) CH
   D) $CH_4$
8) What is the coefficient in front of the $O_2$ when the following reaction is correctly balanced?

   _____$C_3H_8$ + _____ $O_2$ → _____$CO_2$ + _____ $H_2O$

   A) 1 B) 2 C) 3 D) 4 E) 5

# Ied 2nd Semester Final Exam Answers

RM Cervero

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