

Gravimetric Analysis of Metal Carbonate

Experiment Overview

The purpose of this lab is to determine the identity of a Group 1 metal carbonate compound by gravimetric analysis. The unknown is weighed and dissolved in water. A solution of calcium chloride is added to the metal carbonate solution to precipitate the carbonate ions as calcium carbonate. The precipitate is filtered, dried, and weighed. From the data, the formula weight and identity of the unknown metal carbonate is determined. Concepts covered in this lab are double replacement reaction (precipitate formation) and gravimetric analysis.

Background

The identity of group 1 metal M is determined by analyzing an unknown Group 1 metal carbonate, M_2CO_3 .

There are 3 main reactions in this lab:

1. Equation 1: $M_2CO_3 (s) \rightarrow 2M^+ (aq) + CO_3^{2-} (aq)$
2. Equation 2: $Ca^{2+} (aq) + CO_3^{2-} (aq) \rightarrow CaCO_3 (s)$
3. Equation 3: $CaCl_2 (aq) + M_2CO_3 (aq) \rightarrow CaCO_3 (s) + 2MCl (aq)$

The precipitated calcium carbonate is then filtered, dried, and weighed. *The moles of calcium carbonate, $CaCO_3$, are equal to the moles of Group 1 metal carbonate, M_2CO_3 , added to the original solution.* Dividing the mass of the unknown carbonate by the moles of calcium carbonate yields the formula weight, and thus the identity, of the Group 1 metal carbonate.

Pre-Lab Questions: Use the sample data table to answer the following questions:

Sample Data Table:

Mass of crucible + M_2CO_3	12.627 g
Mass of crucible	10.655 g
Mass of M_2CO_3	1.972 g
Mass of filter paper + $CaCO_3$	2.436 g
Mass of the filter paper	0.598 g
Mass of the $CaCO_3$	1.838 g
Moles of $CaCO_3$	mol
Molar mass of M_2CO_3	g/mol
Identity of M_2CO_3	
Percent error	%

1. From the mass of $CaCO_3$, calculate the moles of $CaCO_3$ precipitated.
2. Calculate the molar mass of the unknown.
3. Calculate the molar mass of the following Group 1 metal carbonates:
 - a. Li_2CO_3
 - b. Na_2CO_3
 - c. K_2CO_3
4. What is the identity of M_2CO_3 ?
5. Calculate the percent error in the molar mass determination of M_2CO_3 by comparing the experimentally determined molar mass of M_2CO_3 to the known molar mass of the appropriate metal carbonate.

Gravimetric Analysis Of A Metal Carbonate Answers

Lingsheng Yao



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Table of Contents Gravimetric Analysis Of A Metal Carbonate Answers

1. Understanding the eBook Gravimetric Analysis Of A Metal Carbonate Answers
 - The Rise of Digital Reading Gravimetric Analysis Of A Metal Carbonate Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Gravimetric Analysis Of A Metal Carbonate Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gravimetric Analysis Of A Metal Carbonate Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gravimetric Analysis Of A Metal Carbonate Answers
 - Personalized Recommendations
 - Gravimetric Analysis Of A Metal Carbonate Answers User Reviews and Ratings
 - Gravimetric Analysis Of A Metal Carbonate Answers and Bestseller Lists
5. Accessing Gravimetric Analysis Of A Metal Carbonate Answers Free and Paid eBooks
 - Gravimetric Analysis Of A Metal Carbonate Answers Public Domain eBooks

- Gravimetric Analysis Of A Metal Carbonate Answers eBook Subscription Services
- Gravimetric Analysis Of A Metal Carbonate Answers Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - Gravimetric Analysis Of A Metal Carbonate Answers Compatibility with Devices
 - Gravimetric Analysis Of A Metal Carbonate Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gravimetric Analysis Of A Metal Carbonate Answers
 - Highlighting and Note-Taking Gravimetric Analysis Of A Metal Carbonate Answers
 - Interactive Elements Gravimetric Analysis Of A Metal Carbonate Answers
- 8. Staying Engaged with Gravimetric Analysis Of A Metal Carbonate Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gravimetric Analysis Of A Metal Carbonate Answers
- 9. Balancing eBooks and Physical Books Gravimetric Analysis Of A Metal Carbonate Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gravimetric Analysis Of A Metal Carbonate Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gravimetric Analysis Of A Metal Carbonate Answers
 - Setting Reading Goals Gravimetric Analysis Of A Metal Carbonate Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gravimetric Analysis Of A Metal Carbonate Answers
 - Fact-Checking eBook Content of Gravimetric Analysis Of A Metal Carbonate Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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