

1. What volume of hydrogen gas at STP is produced from the reaction of 50.0g of Mg and 75.0 grams of HCl? How much of the excess reagent is left over (in grams)?



Do a standard mass to mass problem starting with each reactant

50.0 g Mg	$\frac{1 \text{ mol Mg}}{24.31 \text{ g Mg}}$	$\frac{1 \text{ mol H}_2}{1 \text{ mol Mg}}$	$\frac{22.4 \text{ L}}{1 \text{ mol H}_2}$	= 46.1 L H ₂
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75.0 g HCl	$\frac{1 \text{ mol HCl}}{36.46 \text{ g HCl}}$	$\frac{1 \text{ mole H}_2}{2 \text{ mol HCl}}$	$\frac{22.4 \text{ L}}{1 \text{ mol H}_2}$	= 23.0 L H ₂
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HCl is the limiting reactant!!

Flinn Scientific Pogil Activity Limiting And Excess Reactants

Annelies Wilder-Smith



Flinn Scientific Pogil Activity Limiting And Excess Reactants:

Unveiling the Energy of Verbal Beauty: An Mental Sojourn through **Flinn Scientific Pogil Activity Limitting And Excess Reactants**

In a global inundated with screens and the cacophony of quick connection, the profound energy and psychological resonance of verbal beauty often diminish into obscurity, eclipsed by the constant onslaught of noise and distractions. However, set within the lyrical pages of **Flinn Scientific Pogil Activity Limitting And Excess Reactants**, a charming perform of fictional beauty that pulses with natural feelings, lies an remarkable trip waiting to be embarked upon. Penned with a virtuoso wordsmith, that exciting opus guides readers on a mental odyssey, delicately revealing the latent possible and profound impact stuck within the intricate web of language. Within the heart-wrenching expanse with this evocative examination, we will embark upon an introspective exploration of the book is key styles, dissect their charming writing style, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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