#### Module 11 Study Guide Metabolism

Metabolic Pathways

- Several individual reactions
- Catabolism: degradation in which nutrients and cell constituents are broken down to salvage components and make energy
  - Exergonic oxidation of nutrient molecules
  - Free energy released and used to drive endergonic processes
- Anabolism: biosynthesis

Stepwise Oxidation of Metabolites During Glycolysis and Citric Acid Cycle

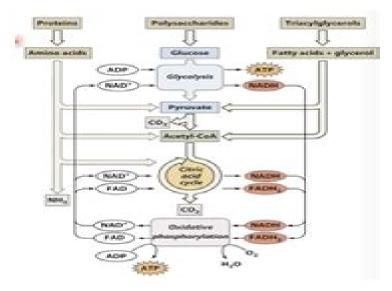
- Reduction of electron carriers NADH and FADH<sub>2</sub>
- NAD' and FAD → Electron carriers
  - NAD : nicotinamide adenine dinucleotide
    - Coenzyme
    - Two electron reduction
  - FAD: flavin adenine dinucleotide
    - One and two electron redox reactions
  - Both must undergo reversible reduction so that they can accept electrons, pass them onto other electron carriers and thereby be regenerated to participate in additional cycles of oxidation and reduction
  - The protein components of redox enzyme play active roles in electron transfer reactions by modulating the reduction potentials of their bound redox active center
- Eventual release of carbon in fully oxidized form: CO<sub>2</sub>
- Production of ATP via substrate-level phosphorylation

Oxidative Phosphorylation

- · Oxidation of electron carriers NADH and FADH2 by ETC
- Passage of electrons down ETC to O<sub>2</sub> → reducing O<sub>2</sub> to H<sub>2</sub>O
  - Cellular respiration at work
  - Oxygen can only accept unpaired electrons because each of its two available lowest energy molecular orbitals is already occupied by one electron
- Creation of H' gradient
- H' gradient coupled to ATP synthesis by ATP synthase
- Overall goal is to make ATP

Generalizations

- Catabolic Pathways
  - Extract energy by oxidizing fuel molecules— input for biological macromolecules forced into the same process
  - Converge on a few intermediates (ex: Coenzyme A→ acetyl coenzyme A)
  - Generate ATP
  - May generate reducing power and carbon skeletons by biosynthesis
    - Reduced electron carriers



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Robin R. Preston, Thad E. Wilson

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