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Page 8: Tables and Figures

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Abstract

1. What effect is meant by the term "binding energy" in the context of a nucleus?
2. Is there a difference between the binding energy of a nucleus and the binding energy of a nucleon?
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Chemistry Education in the ICT Age Minu Gupta Bhowon, Sabina Jhaumeer-Laulloo, Henri Li Kam Wah, Ponnadurai Ramasami, 2009-07-21. The 20 International Conference on Chemical Education (ICCE) which had Chemistry in the ICT Age as the theme was held from 3 to 8 August 2008 at Le Meridien Hotel Pointe aux Piments in Mauritius. With more than 200 participants from 40 countries, the conference featured 140 oral and 50 poster presentations. Participants of the 20 ICCE were invited to submit full papers and the latter were subjected to peer review. The selected accepted papers are collected in this book of proceedings. This book of proceedings encloses 39 presentations covering topics ranging from fundamental to applied chemistry such as Arts and Chemistry Education, Biochemistry and Biotechnology, Chemical Education for Development, Chemistry at Secondary Level, Chemistry at Tertiary Level, Chemistry Teacher Education, Chemistry and Society, Chemistry Olympiad, Context Oriented Chemistry, ICT and Chemistry Education, Green Chemistry, Micro Scale Chemistry, Modern Technologies in Chemistry Education, Network for Chemistry and Chemical Engineering Education, Public Understanding of Chemistry, Research in Chemistry Education and Science Education at Elementary Level. We would like to thank those who submitted the full papers and the reviewers for their timely help in assessing the papers for publication. We would also like to pay a special tribute to all the sponsors of the 20 ICCE and in particular the Tertiary Education Commission (<http://tec.intnet.mu>) and the Organisation for the Prohibition of Chemical Weapons (<http://www.opcw.org>) for kindly agreeing to fund the publication of these proceedings.

Limiting Reactant and Percent Yield Lifeliqe, 2019. This lesson plan covers analyzing chemical reactions in order to determine limiting reactants and excess reactants including calculating the amount of excess reactant, calculate the theoretical yield of a reaction and calculating the percent yield of a reaction.

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