STOICHIOMETRY: LIMITING REAGENT

Name _____

1. $N_2 + 3H_2 \rightarrow 2NH_3$

How many grams of $\mathrm{NH_3}$ can be produced from the reaction of 28 g of $\mathrm{N_2}$ and 25 g of $\mathrm{H_3}$?

2. How much of the excess reagent in Problem 1 is left over?

Mg + 2HCl → MgCl₂ + H₂

What volume of hydrogen at STP is produced from the reaction of $50.0~{\rm g}$ of Mg and the equivalent of 75 g of HCI?

4. How much of the excess reagent in Problem 3 is left over?

3AgNO₃ + Na₃PO₄ → Ag₃PO₄ + 3NaNO₃

Silver nitrate and sodium phosphate are reacted in equal amounts of 200. g each. How many grams of silver phosphate are produced?

6. How much of the excess reagent in Problem 5 is left?

Limiting Reagents Answer Key

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Limiting Reagents Answer Key:

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