

Colorimetry (Jan 2010) 8 marks

1. use filter of complementary colour to iodine/solution
 2. zero colorimeter with water
 3. measure absorbance readings of standard solutions/solutions of known concentration (of iodine)
 4. plot calibration graph
 5. take absorbance readings of the reacting mixture at known/certain times AND (must refer to time – may state units of time)
 6. convert absorbance readings to iodine concentrations using the calibration curve
- OR/AND:

In order to gain the mark for point 3, 5 or 6, absorbance must be used AND spelled correctly at least once

ALLOW

- ☐ filter/wavelength giving maximum absorption/absorbance
- ☐ green/blue filter
- ☐ suitable filter

orange/yellow/brown is OK

is using starch ANYWHERE is a CON and CANNOT gain marking point 3

NOT 'solvent' instead of 'water'

CONFOR reference to quenching procedures

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JA Banks



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