

Intro to Acids and Bases

Name: _____
 Class: _____
 Date: _____

Define each term: Definitions of acids and bases:

Arrhenius Definitions:

- acid substances that release dissolved hydrogen ions (H^+) (acids H^+ gases formed in gases as the hydrogens form H_2 and H_2O (acid H_2O) and conjugated anions)
- base substances which release hydroxide ions (OH^-)

Brønsted-Lowry Definitions:

- acid substances that donate a proton (H^+) (acid donate)
- base substances which accept a proton (H^+) (base receive)

Table 1: Representative reactions for H^+ (acid) react with base representative reactions:

Reaction	Eq.
1. $HCl(aq) + NaOH(aq) \rightarrow NaCl(aq) + H_2O(l)$	$H^+(aq) + OH^-(aq)$
2. $HNO_3(aq) + NaOH(aq) \rightarrow NaNO_3(aq) + H_2O(l)$	$H^+(aq) + OH^-(aq)$
3. $H_2SO_4(aq) + NaOH(aq) \rightarrow NaHSO_4(aq) + H_2O(l)$	$H^+(aq) + OH^-(aq)$
4. $H_2SO_4(aq) + NaOH(aq) \rightarrow Na_2SO_4(aq) + 2H_2O(l)$	$2H^+(aq) + 2OH^-(aq)$
5. $HCl(g) + NaOH(s) \rightarrow NaCl(s) + H_2O(l)$	$H^+(g) + OH^-(s)$

Critical Thinking Questions

- Are all of the reactions in Table 1 that are representative acids?
 (a) no all some none
- Are all of the products in Table 1 that are representative bases?

Class work

- Are all of the products in Table 1 that are Brønsted-Lowry acids?

Individual questions for students:

- Are all of the products in Table 1 that are Brønsted-Lowry bases?

Group questions:

- Are products that are not an acid and base? Explain.

Discussion and class work:

- Class members write on their own sheets how they define different conditions? Explain.

Homework

- Write your answers: "Are substances that are strong acids and bases Brønsted-Lowry acids?" Explain.

Intro To Acids And Bases Chemquest 48

Ying Liu



Intro To Acids And Bases Chemquest 48:

The Enigmatic Realm of **Intro To Acids And Bases Chemquest 48**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing short of extraordinary. Within the captivating pages of **Intro To Acids And Bases Chemquest 48** a literary masterpiece penned by a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting affect the hearts and minds of those who partake in its reading experience.

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