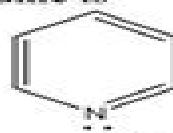


12. Hybridisation of nitrogen atom in pyridine is



1. sp^3 2. sp^2 3. sp 4. sp^3d
13. The correct IUPAC name of the following compound is:
- 
1. 5, 6-diethyl-8-methyldec-6-ene 2. 5,6-diethyl-3-methyloct-4-ene
3. 5,6-diethyl-3-methyldec-4-ene 4. 2,4,5-triethylnon-3-ene
14. The ratio of pure and hybriide orbitals is $H_2C = CH - CH = CH_2$
1. 7 : 12 2. 14 : 13 3. 12 : 10 4. 5 : 6
15. Which of the following types of carbon atom possess highest electro negativity
1. sp^3 2. sp^2 3. sp 4. All have equal E.N
16.
$$\begin{array}{c} CH_3CH_2CH_2CH_2CH_2CHCH_2-CH-C_2H_5 \\ | \\ CH_2 \\ | \\ CH_3-C-CH_3 \\ | \end{array}$$

In the above compound, the number of secondary carbon atoms present in the above compound are respectively:

- [illegible]

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