

Precipitation Reactions

- Reactions that produce insoluble products (or precipitates) when two aqueous solutions are mixed.
- Examples:
- (1) $\text{AgNO}_3(aq) + \text{NaCl}(aq) \rightarrow \text{AgCl}(s) + \text{NaNO}_3(aq)$
- (2) $\text{Pb}(\text{NO}_3)_2(aq) + \text{K}_2\text{CrO}_4(aq) \rightarrow \text{PbCrO}_4(s) + 2\text{KNO}_3(aq)$
- (3) $\text{BaCl}_2(aq) + \text{H}_2\text{SO}_4(aq) \rightarrow \text{BaSO}_4(s) + 2\text{HCl}(aq)$
- (4) $3\text{AgNO}_3(aq) + \text{Na}_3\text{PO}_4(aq) \rightarrow \text{Ag}_3\text{PO}_4(s) + 3\text{NaNO}_3(aq)$

Experiment 17 Precipitation Reactions Answers

Jeffrey J. W. Baker



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 16 Sustainable Management of Natural Resoures Practical Project Appendix Answer Sheet Examination Paper MCQs in

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