

## Naming Review Practice

Directions: On the left of each question, identify if the compound/molecule is Ionic (write an I) or Covalent (write a C). Then, either write the chemical name of the molecule (if given the formula) or the chemical formula of the molecule (if given the name)

- |                 |                                 |                                       |
|-----------------|---------------------------------|---------------------------------------|
| <b>Covalent</b> | 1. $\text{CS}_3$                | Carbon Trisulfide                     |
| <b>Ionic</b>    | 2. $\text{TiO}_2$               | Titanium IV Oxide                     |
| <b>Ionic</b>    | 3. $\text{CaCO}_3$              | Calcium Carbonate                     |
| <b>Ionic</b>    | 4. $\text{Cu}(\text{NO}_2)_2$   | Copper II Nitrite                     |
| <b>Ionic</b>    | 5. $\text{Sn}(\text{CN})_4$     | Tin IV Cyanide                        |
| <b>Ionic</b>    | 6. $\text{Ag}_2\text{SO}_3$     | Silver Sulfite                        |
| <b>Ionic</b>    | 7. $\text{NiSe}_2$              | Nickel IV Selenide                    |
| <b>Covalent</b> | 8. $\text{P}_2\text{O}_5$       | Triphosphorus Pentoxide               |
| <b>Ionic</b>    | 9. $\text{Sn}_3(\text{PO}_4)_2$ | Tin II Phosphate                      |
| <b>Covalent</b> | 10. $\text{N}_7\text{Cl}_3$     | Heptanitrogen Trichloride             |
| <b>Ionic</b>    | 11. Manganese (III) bromide     | $\text{MnBr}_3$                       |
| <b>Covalent</b> | 12. Sulfur dioxide              | $\text{SO}_2$                         |
| <b>Ionic</b>    | 13. Tin (IV) sulfate            | $\text{Sn}(\text{SO}_4)_2$            |
| <b>Ionic</b>    | 14. Zinc hydroxide              | $\text{Zn}(\text{OH})_2$              |
| <b>Covalent</b> | 15. Carbon tetrachloride        | $\text{CCl}_4$                        |
| <b>Ionic</b>    | 16. Barium cyanide              | $\text{Ba}(\text{CN})_2$              |
| <b>Ionic</b>    | 17. Cobalt (VI) Chromate        | $\text{Co}(\text{CrO}_4)_3$           |
| <b>Ionic</b>    | 18. Lithium nitrite             | $\text{LiNO}_2$                       |
| <b>Covalent</b> | 19. Trisulfur hexafluoride      | $\text{S}_3\text{F}_6$                |
| <b>Ionic</b>    | 20. Aluminum oxalate            | $\text{Al}_2(\text{C}_2\text{O}_4)_3$ |

# Naming Review Guide Answer Key

**N Noddings**



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