

A Microscale Titration

Propose a series of microscale titrations involving the formation of a salt between a weak acid and a weak base. Record the observed color change.

Observe color changes.

1. Propose a series of microscale titrations between a:

Weak acid: acetic acid

Weak base: ammonia

Observe 20 drops:

2. After about 5, the color of the solution is:

yellow to orange

3. Repeat the titration from Procedures 1, 2, and 3, using:

Weak acid: acetic

Weak base: ammonia

Observe 20 drops:

4. What do you think is the color of the average compound that you made?

Observed color: yellow to orange

5. What does the color of the solution tell you about the acidity or basicity of the solution, and the compound that you made? (Remember, indicators react with acids and bases differently.)

Reaction of acid with the observed compound:

6. How does the color change tell you that the reaction is an acid-base reaction?

Reaction of indicator:

Microscale Titration Lab Answers

Raffaella Di Napoli



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