Class	General Formula	Example	Common Name (Systematic Name)	Common Suffix/Prefix (Systematic)
		Hydrocarbons		
Alkanes	RH	CH ₃ CH ₃	ethane	-ane
Alkenes	RR'C-CR"R"	H_2C — CH_2	ethylene (ethene)	-ene
Alkynes	RC=CR'	нс—сн	acetylene (ethyne)	(-yne)
Arenes	ArH°		benzene	-ene
	Hale	ogen-Containing Comp	ounds	
Alkyl halides	RX	CH ₃ CH ₂ CI	ethyl chloride (chloroethane)	halide (halo-)
Aryl halides	ArX"		chlorobenzene	halo-
	Oxy	gen-Containing Compo	ounds	
Alcohols	ROH°	CH ₃ CH ₂ OH	ethyl alcohol (ethanol)	-ol
Phenols	ArOH*	>он	phenol	-ol
Ethers	ROR"	H ₃ CH ₂ COCH ₂ CH ₃	diethyl ether	ether
Aldehydes	RCHO	CH ₃ CH	acetaldehyde (ethanal)	-aldehyde (-al)
Ketones	RR'C=O	CH ₃ CCH ₃	acetone (2-propanone)	-one
Carboxylic acids	RCO₂H	CH ₃ COH	acetic acid (ethanoic acid)	-ic acid (-oic acid)
		arboxylic Acid Derivati	ves	
Esters	RCO₂R*	CH ₃ COCH ₃	methyl acetate (methyl ethanoate)	-ate (-oate)
Amides	RCONHR'	CH ₃ CNHCH ₃	N-methylacetamide	-amide
	Nitr	ogen-Containing Comp	ounds	
Amines	RNH ₂ , RNHR", RNR'R"	CH ₃ CH ₂ NH ₂	ethylamine	-amine
Nitriles	RC≡N	H ₃ CC≡ N	acetonitrile	-nitrile
Nitro compounds	ArNO2"	NO ₂	nitrobenzene	nitro-
R indicates an alkyl gro	up ^b Ar indicates an <i>aryl</i>	group.		

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Table of Contents Naming Compounds Cheat Sheet

- 1. Understanding the eBook Naming Compounds Cheat Sheet
 - The Rise of Digital Reading Naming Compounds Cheat Sheet
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Naming Compounds Cheat Sheet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Naming Compounds Cheat Sheet
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Naming Compounds Cheat Sheet
 - Personalized Recommendations
 - Naming Compounds Cheat Sheet User Reviews and Ratings
 - Naming Compounds Cheat Sheet and Bestseller Lists
- 5. Accessing Naming Compounds Cheat Sheet Free and Paid eBooks

- Naming Compounds Cheat Sheet Public Domain eBooks
- Naming Compounds Cheat Sheet eBook Subscription Services
- Naming Compounds Cheat Sheet Budget-Friendly Options
- 6. Navigating Naming Compounds Cheat Sheet eBook Formats
 - o ePub, PDF, MOBI, and More
 - Naming Compounds Cheat Sheet Compatibility with Devices
 - Naming Compounds Cheat Sheet Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Naming Compounds Cheat Sheet
 - Highlighting and Note-Taking Naming Compounds Cheat Sheet
 - Interactive Elements Naming Compounds Cheat Sheet
- 8. Staying Engaged with Naming Compounds Cheat Sheet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Naming Compounds Cheat Sheet
- 9. Balancing eBooks and Physical Books Naming Compounds Cheat Sheet
 - Benefits of a Digital Library
 - \circ Creating a Diverse Reading Collection Naming Compounds Cheat Sheet
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Naming Compounds Cheat Sheet
 - Setting Reading Goals Naming Compounds Cheat Sheet
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Naming Compounds Cheat Sheet
 - Fact-Checking eBook Content of Naming Compounds Cheat Sheet
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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