

# Experiment 5C: Types of Chemical Reactions

## Objectives:

- To observe different chemical reactions
- To interpret and explain observations with balanced chemical equations
- To identify reactions from an input of their main types

**Materials:** Please refer to Experiment 5C and page 7 of the General Chemistry Laboratory Experiment 5C Manual

## Procedures:

1. Check any safety equipment such as goggles and aprons.
2. Make observations, before heating, and after each reaction. Observe and describe in Table 1.

### Reaction 1:

1. Ignite burner flame for heat.
2. Using crucible tongs, hold a pre-weighed cylinder of copper up to the hottest part of the burner as heat is applied.

### Reaction 2:

1. Clean silver nail with a piece of steel wool to remove the oxide layer of the nail coating.
2. Place the nail in a test tube and add a copper (II) sulfate solution and heat the entire mixture.
3. After approximately 10 minutes, remove the test tube and any changes in reaction heat and the solution colour (see a description below's results).

### Reaction 3:

1. Place some solid copper (I) chloride pentahydrate in a test tube and heat it evenly until dry.
2. Using crucible tongs, hold the test solution contents in a test tube using crucible tongs and observe. Heat the test tube in a crucible tongs and crucible tongs in a crucible flame.
3. Continue heating until no further change is observed. Refer to Reaction 1.

# Lab Experiment 5c Answers

**Daniel F McAuley**



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