

**Question 2****[5]**

You are provided with two organic compounds, C-16 and C-17.

Perform the experiments given below on each of the two compounds. Record the changes taking place at every step of the experiment.

Note the smell of the substance formed, if significant, the colour of the solution obtained, the colour of the precipitate produced and any other observations you may have. State the identity of each compound on the basis of the experiments and observational changes.

**PROCEDURE:****(a) Substance C – 16**

- (i) Take 1 ml of C-16 in a test tube and add 1 ml of Fehling's solution. Warm the contents.
- (ii) Take 2 ml of C-16 in a test tube and add 1 ml of lead acetate solution followed by 1 ml of ammonium hydroxide solution. Boil the contents.
- (iii) Take 2 ml of C-16 in a test tube and add 2 to 3 drops of alcoholic  $\alpha$ -naphthol solution. Pour conc. sulphuric acid slowly along the side of the test-tube.

Show the results as required to the Visiting Examiner.

**(b) Substance C-17**

- (i) Take 1 ml of C-17 in a test-tube and add a few drops of water. Now add about 1 ml of sodium hypochlorite solution; Shake the contents.
- (ii) Take 1 ml of C-17 in a test-tube and add a few drops of dilute sulphuric acid. Now add 1 ml of potassium dichromate solution. Shake and warm the contents.
- (iii) Take 1 ml of C-17 in a test-tube and add 1 ml of conc. hydrochloric acid to it. Now, add a few drops of neutral ferric chloride solution and dilute the contents with water.
- (iv) Take 2 ml of C-17 and dilute it with water. To this, add a few drops of bleaching powder solution and shake. Now, add a few drops of ammonium sulphide solution and shake.

Show the results as required to the Visiting Examiner.

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