

Name _____

Date _____

Period _____

VIDEO GUIDE: NOVA – “Hunting the Elements”

PERIODIC TABLE OF THE ELEMENTS

- 1) How many different elements are there? _____
- 2) What is the symbol for gold? _____
- 3) What is gold's atomic number? _____
- 4) What are some unique properties of gold? _____
- 5) What process is used to separate gold atoms from crushed rock? _____
- 6) How much gold is harvested per ton of earth from the underground mine? _____



7) Gold does not share its electrons easily. Therefore, it is not very chemically reactive. What are two other metals that, like gold, don't react with other elements? _____

8) What is the worth (\$\$) of one 60-pound bar of gold? _____

9) Two rows above gold is the metal: _____; its symbol is: _____

10) In what objects / industries is copper used? _____

11) What are some unique properties of copper? _____

12) What is the symbol for tin? _____ What is its atomic number? _____

13) When copper and tin are mixed together, the alloy _____ is formed.

14) Why isn't copper alone used to make bells? _____



15) It is possible that the crack in the famous Liberty Bell was caused by too much _____

16) Metals form crystals, which means that they have an _____ of atoms.

17) Atoms with 1 proton are the element: _____; atoms with 2 protons are _____.

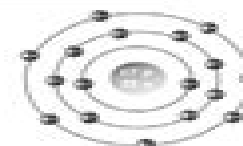
18) The number of protons in an atom is known as that element's _____.

19) The chemical symbol of each element is based on the element's name in which language? _____

20) More than 70% of the elements on the periodic table are _____.

21) Why are the noble gases called "noble"? _____

22) Which subatomic particles determine an element's reactivity? (circle correct answer)

PROTONS**NEUTRONS****ELECTRONS**

23) How many electrons are in the first shell of electrons? _____ The second? _____

24) How many electrons do the noble gases have in their outermost shell? _____

25) How many electrons does chlorine have in its outermost shell? _____

26) What will an atom of chlorine do to complete its outer shell? _____

27) The alkali metals, such as sodium, have _____ electron in their outer shell.

28) What do the alkali metals do with that one electron? _____



Hunting The Elements Nova Worksheet Answers

Karin Nielsen-Saines



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Table of Contents Hunting The Elements Nova Worksheet Answers

1. Understanding the eBook Hunting The Elements Nova Worksheet Answers
 - The Rise of Digital Reading Hunting The Elements Nova Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Hunting The Elements Nova Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Hunting The Elements Nova Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Hunting The Elements Nova Worksheet Answers

- Personalized Recommendations
- Hunting The Elements Nova Worksheet Answers User Reviews and Ratings
- Hunting The Elements Nova Worksheet Answers and Bestseller Lists
- 5. Accessing Hunting The Elements Nova Worksheet Answers Free and Paid eBooks
 - Hunting The Elements Nova Worksheet Answers Public Domain eBooks
 - Hunting The Elements Nova Worksheet Answers eBook Subscription Services
 - Hunting The Elements Nova Worksheet Answers Budget-Friendly Options
- 6. Navigating Hunting The Elements Nova Worksheet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Hunting The Elements Nova Worksheet Answers Compatibility with Devices
 - Hunting The Elements Nova Worksheet Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Hunting The Elements Nova Worksheet Answers
 - Highlighting and Note-Taking Hunting The Elements Nova Worksheet Answers
 - Interactive Elements Hunting The Elements Nova Worksheet Answers
- 8. Staying Engaged with Hunting The Elements Nova Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Hunting The Elements Nova Worksheet Answers
- 9. Balancing eBooks and Physical Books Hunting The Elements Nova Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Hunting The Elements Nova Worksheet Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Hunting The Elements Nova Worksheet Answers
 - Setting Reading Goals Hunting The Elements Nova Worksheet Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Hunting The Elements Nova Worksheet Answers

- Fact-Checking eBook Content of Hunting The Elements Nova Worksheet Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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