

Analytical Chemistry MCQs: TYBSC SEM III

1] Gravimetric Analysis

1. If the solubility of calcium fluoride in water is S moles per litre, its solubility is given by

- (a) $2S$ (b) $2S^2$
(c) $4S^3$ (d) $4S^2$

2. Correct representation of solubility product of $\text{Ca}_3(\text{PO}_4)_2$ is

- (a) $[\text{Ca}^{2+}][\text{PO}_4^{3-}]^2$ (b) $[\text{Ca}^{2+}][\text{PO}_4^{3-}]$
(c) $[\text{Ca}][\text{PO}_4]$ (d) none of these

3. Dimethyl glyoxime is a reagent specific for the metal ion.

- (a) Al (b) Ni
(c) Fe (d) Cu

4. The solubility product expression for Hg_2Cl_2 is

- (a) $[\text{Hg}^{2+}][\text{Cl}^-]$ (b) $[\text{Hg}_2^{2+}][\text{Cl}^-]^2$
(c) $[\text{Hg}_2^{2+}][2\text{Cl}^-]$ (d) $[\text{Hg}^{+1}][\text{Cl}^-]$

5. A solution is called saturated if

- (a) ionic product < solubility product.
(b) ionic product > solubility product.
(c) ionic product = solubility product.
(d) none of these

6. Solubility of alcohol in water is due to

- (a) ionic bonding
(b) covalent bonding
(c) Metallic bonding
(d) Hydrogen bonding

7. Precipitation takes place when the product of concentration of ions

- (a) Exceeds their solubility product
(b) Is less than their solubility product
(c) Equal to their solubility product
(d) None of these

Mcq Of Analytical Chemistry

R Bogdan



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MCQs Chemistry V. K. Ahluwalia, 2013-12-30 This book has been divided in 37 chapters under three parts Inorganic Chemistry Organic Chemistry and Physical Chemistry for convenient understanding It also includes solved model test papers of the previous three years of AIIMS CBSE PMT CPMT UP to enable students to develop the skills of problem solving and time management essential for any entrance examination In addition to providing answers to all the questions detailed explanatory notes to selected difficult questions have also been provided to justify the answer A separate section of Assertions and Reasons is also given at the end of each chapter Exhaustive Question Bank Explanatory Notes and Hints Assertions Reasons Includes Pre solved papers of five years Models Test Papers of AIIMS CBSE PMT CPMT *Chemistry MCQs for JEE - II* Mr. Ajeet Singh, 2018-05-22 This volume continues the JEE chemistry MCQ series focusing on advanced topics and challenging questions to enhance exam readiness for engineering entrance tests *Class 11-12 Chemistry MCQ (Multiple Choice Questions)* Arshad Iqbal, 2019-05-17 The Class 11 12 Chemistry Multiple Choice Questions MCQ Quiz with Answers PDF College Chemistry MCQ PDF Download Quiz Questions Chapter 1 6 Practice Tests with Answer Key 11th 12th Grade Chemistry Questions Bank MCQs Notes includes revision guide for problem solving with hundreds of solved MCQs Class 11 12 Chemistry MCQ with Answers PDF book covers basic concepts analytical and practical assessment tests Class 11 12 Chemistry MCQ PDF book helps to practice test questions from exam prep notes The Class 11 12 Chemistry MCQs with Answers PDF eBook includes revision guide with verbal quantitative and analytical past papers solved MCQs Class 11 12 Chemistry Multiple Choice Questions and Answers MCQs PDF Free download chapter 1 a book covers solved quiz questions and answers on chapters atomic structure basic chemistry chemical bonding chemistry experimental techniques gases liquids and solids tests for college and university revision guide Class 11 12 Chemistry Quiz Questions and Answers PDF free download eBook s sample covers beginner s solved questions textbook s study notes to practice online tests The book Grade 11 12 Chemistry MCQs Chapter 1 6 PDF includes college question papers to review practice tests for exams Class 11 12 Chemistry Multiple Choice Questions MCQ with Answers PDF digital edition eBook a study guide with textbook chapters tests for NEET MCAT GRE GMAT SAT ACT competitive exam College Chemistry Mock Tests Chapter 1 6 eBook covers problem solving exam tests from chemistry textbook and practical eBook chapter wise as Chapter 1 Atomic Structure MCQ Chapter 2 Basic Chemistry MCQ Chapter 3 Chemical Bonding MCQ Chapter 4 Experimental Techniques MCQ Chapter 5 Gases MCQ Chapter 6 Liquids and Solids MCQ The Atomic Structure MCQ PDF e Book Chapter 1 practice test to solve MCQ questions on Atoms atomic spectrum atomic absorption spectrum atomic emission spectrum molecules azimuthal quantum number Bohr s model Bohr s atomic model defects charge to mass ratio of electron discovery of electron discovery of neutron discovery of proton dual nature of matter electron charge electron distribution electron radius and energy derivation electron velocity electronic configuration of elements energy of revolving electron fundamental particles Heisenberg s uncertainty

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