

### ক-বিভাগ (পাঠগণিত)

১। একই হার মুনাফার কানো মূলধনের এক বছরে চ র্কা ৬৫০ টাকা এবং দুই বছরে চ র্কা ৬৭৬ টাকা

ক) উদ্দীপকের তথ্যগুলোকে দুটি সমীকরণে প্রকাশ করো ২

খ) ই বছরে সরল মুনাফা ও চ র্কা মুনাফার পার্থক্য নির্ণয় করো ৪

গ) কত বছরে ওই পরিমাণ মূলধন ওই নির্দিষ্ট হার মুনাফায় তিন গুণ হবে? ৪

২। একটি বগক্ষেত্র ও একটি আয়তক্ষেত্রে পরিসীমা সমান আয়তক্ষেত্রে দৈর্ঘ্য প্রস্থের তিন গুণ এবং ক্ষেত্রফল ৩০৭২ বর্গমিটার বগক্ষেত্রটির ঠিক মাঝ বরাবর আড়াআড়িভাবে ৪ মিটার প্রশস্ত দুটি পথ আছে ৫০ সেমি. দৈর্ঘ্য ও ৪০ সেমি. প্রস্থের ইট দিয়ে রাস্তা দুটি বাঁধা করা হলো ইটের হাজার ৭৫০০ টাকা

ক) আয়তক্ষেত্রে প্রস্থ  $x$  মিটার ধরে আয়তক্ষেত্রে ক্ষেত্রফলকে একটি দ্বারা প্রকাশ করো ২

খ) বগক্ষেত্রে ক্ষেত্রফল নির্ণয় করো ৪

গ) রাস্তা দুটি বাঁধাই করতে কত টাকার ইট লাগবে? ৪

### খ-বিভাগ (বীজগণিত)

৩।  $x + y = 13$  এবং  $x - y = 3$

ক)  $x^2 - y^2$  এর মান নির্ণয় করো ২

খ)  $2(x^2 + y^2)$  এর মান নির্ণয় করো ৪

গ) দেখাও যে,  $x^2 - y^2 - xy = -1$ .

৪।  $8x + y = p$ ;  $ax - by = q$

ক)  $pb + q$  এর মান কত? ২

খ)  $p = a - b$  এবং  $q = a^2 + b^2$  হলে  $(x, y)$  নির্ণয় করো ৪

গ)  $p/q = 1/2$  হলে  $x/y$  এর মান নির্ণয় করো ৪

৫।  $U = \{1, 2, 3, 4, 5, 6\}$ ,  $A = \{1, 3, 5\}$ ,  $B = \{2, 4, 6\}$  এবং  $C = \{1, 6\}$

ক)  $U - C$  নির্ণয় করো ২

খ) দেখাও যে,  $(A \cap B) \cap C = A \cap (B \cap C)$

গ) দেখাও যে,  $A \cap (B \cap C) = (A \cap B) \cap (A \cap C)$

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**DJ Losen**



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web write the oxidation and reduction reactions for each redox reaction the first one is done for you  $23 \text{ Fe} + \text{Co} \rightarrow \text{Co}^{2+} + \text{Fe}$   
oxidation  $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^-$  reduction  $\text{Fe}^{2+} + 2\text{e}^- \rightarrow \text{Fe}$   $24 \text{ 3 Ag} + \text{Ni} \rightarrow \text{Ni}^{3+} + 3 \text{ Ag}$  oxidation reduction  $25 \text{ Cu} + 2 \text{ Pb} \rightarrow \text{Pb}^{2+} + \text{Cu}$  oxidation reduction  $26 \text{ O}_2$

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web in each of the following redox reactions indicate the elements that undergo changes in oxidation number and the magnitude of change then identify the element oxidized and reduced as well as the oxidizing agent and reducing agent a  $\text{I}_2 + \text{O}_5$

5co i2 5co2 i 2 o 5 5 c o i 2 5 c o 2

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web chemical reactions in which there is a transfer of electrons from one substance to another are known as oxidation reduction reactions or redox reactions in this unit you will examine the oxidation reduction process and use the oxidation state and oxidation number concepts to identify redox reactions and to keep track of electrons

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